### Module Handbook

### Master of Science Biomedical Chemistry

## **Contents / Modules**

Module descriptions	
Mandatory Modules	
Biochemistry	3
Practical Course in Molecular Biology and Biochemistry	5
Organic Chemistry	7
Practical Course on Molecule Synthesis	
Pharmacology for Natural Scientists	
Pharmaceutical Sciences for Natural Scientists	11
Research Project	13
Master Thesis	
Elective Area	15
Chemistry of Natural Products	15
Radiopharmaceutical Chemistry	17
Practical Course Selected Aspects of Medicinal Chemistry	18
Bioinorganic Chemistry	20
Biophysical Chemistry	
Toxicology 1	22
Toxicology 2	
Immunological Principles	25
Practical Exercises in Immunology	
Pharmaceutical Biology	
Practical Course in Pharamaceutical Biology	
Microbiology and Biotechnology	29
Animal Physiology	
Plant Physiology	
Electrochemistry	33
Integrated Analytical-Preparative Lab Course	
Electrons in Molecules	35
Supramolecular Catalysis	
Molecular Photochemistry	
Advanced Laboratory Course on Functional Molecular Materials	
Trace Analysis I	
Trace Analysis II	40
Macromolecular Chemistry	
Practical Course Biomacromolecular Chemistry	

Colloid Chemistry and Medical Polymers Complex (Supra)Molecular Systems and Biopolymers	
Complex (Supra)Molecular Systems and Biopolymers	47
Modern Methods of Physical Chemistry	
Practical Course Modern Methods of Spectroscopy and Microscopy	50
Introduction in Nuclear Chemistry	51
Lab Course Nuclear Chemistry 1	53
Principles of Quantum Chemistry	54
Programming in Quantum Chemistry	
Practical Computational Chemistry	
Contemporary Topics of Theoretical Chemistry	
Remarks	
Abbreviations	59
ithout	
hormation	

Module Handbook	Version	As of	JGU
Master of Science Biomedical Chemistry	2.0	14.03.2024	JGIO

# Module descriptions

## Mandatory Modules

Module BCF	Biochei	mistry			[Modul-K	ennnummer]
Mandatory or elective Module	M (2nd C	ourse as elective N	lodule)		<u>_</u>	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semester					
Courses/ Learning formats	Туре	Regel term When starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture "Molecular and Cellular Biochemistry"	L	2 (1)	Elective	4	138 h	6
or					0	
b) Lecture "Methods of Biochemistry"	L	1 (2)	Elective	2	69 h	3
c) Supporting seminar to b)	S	1 (2)	Elective	2 📿	69 h	3
In order to complete the module, yo	u have to	fulfil the following	requirements:			
Compulsory Attendance			(	20		
Active participation	According	g to § 5 para. 3	X	V		
Coursework						
Module examination	Usually w a) or of b	ritten exam (120 m ) and c)	in), alternativel	y oral exam (30	0 min) about th	e contents of
Qualification Goals, learning outcom	e, compe	tences				
Students are able to, a) - reproduce essential contents of cell	ula a bia ab			to d fields		
<ul> <li>explain and evaluate principles of get</li> <li>Evaluate the opportunities and risks their own work</li> <li>Assign and explain the principles of</li> <li>to understand and reproduce the bit or use relevant technical terms of cet</li> <li>to critically evaluate the factual kno primary literature published in intern b) and c)</li> <li>to assign appropriate methods to prior to be able to analyse typical data of</li> </ul>	ene regula of geneti signal trar ochemical Ilular bioc wledge co ational sp oblems in	tion and genetic en c engineering, devel and cell biological l chemistry correctly vered in biochemica ecialist journals the fields of protein thods	gineering exper op their own po basics of structu al, cellular and r	iments oint of view an ure-giving proc nolecular biolo	esses ogy textbooks a	

Module Handbook Master of Science Biomedical Chemistry Version As of 2.0 14.03.2024



a)							
Mechanisms of cellular signal transduction, signalling pa	thways, receptors, genome						
<ul> <li>transcriptional regulation, epigenetics, stem cells</li> </ul>							
• Gene transfer in cells and organisms, plasmids, phages,	transfection methods; expression systems						
RNA structures, ribozyme, spliceosome, RNAses, riboswitches							
<ul> <li>Innate and adaptive immunity, haematopoiesis, phagoc</li> </ul>							
<ul> <li>B- and T-cell receptors, cytokines, immunoglobulins, MF</li> </ul>	B- and T-cell receptors, cytokines, immunoglobulins, MHC, monoclonal antibodies, autoimmunity						
	receptors, membrane domains, caveolae, ligand binding, G protein coupled receptors (GPCR), arrestins						
<ul> <li>GPCR-associated diseases, heterotrimeric G proteins, sig</li> </ul>							
<ul> <li>Second messengers (cAMP, cGMP, Ca2+, NO, inositol ph</li> </ul>							
Protein kinase families, PKA, PKC, calmodulin, CaM kinas							
Receptor tyrosine kinases, growth factors, cytokine rece							
Ras family, MAP kinases, regulated proteolysis, secretas							
Nucleolar receptors (steroid receptors, retinoid X recept							
Membrane transport, signal sequences, translocation to							
Protein modifications, unfolded protein response, secre							
<ul> <li>Cytoskeleton (microtubules, actin, intermediate filamen</li> <li>Cell-cell, cell-matrix connections, extracellular matrix, ce</li> </ul>							
<ul> <li>Cell cycle and apoptosis: cyclins, CDKs, IAPs, Bcl proteins</li> <li>Neuronal signal transduction: basics in electrophysiolog</li> </ul>							
	y, ion channels						
b) and c)							
<ul> <li>Methods of protein expression</li> <li>Principles and methods of protein isolation and identific</li> </ul>	ation						
<ul> <li>Principles and methods of protein isolation and identific</li> <li>Immune techniques in biochemistry</li> </ul>	ation						
<ul> <li>Spectroscopic methods in biochemistry</li> </ul>							
<ul> <li>Methods of protein structure analysis</li> </ul>							
<ul> <li>Protein stability</li> </ul>							
Protein dynamics							
Chemical modification of proteins							
Biochemistry and biophysics of lipid membranes							
Membrane proteins							
<ul> <li>In vivo and in vitro studies of protein-protein and protei</li> </ul>	n-lipid interactions						
Microscopic techniques							
Expression and protein characterisation in vivo							
Compulsory entrance requirements							
Recommended participation requirement(s) for the module							
and/or individual courses of the module							
Language(s) of instruction and examination	German or English						
Weight of the module grade in the overall grade	12/66; as elective: not graded						
	a) Only in the summer term						
Frequency of module offer	b), c) Only in the winter term						
Reasons for compulsory attendance							
Person responsible for the module	UnivProf. Dr. Dirk Schneider						
	Bachelor of Science Molecular Biotechnology, Master of						
Transferability of the module to other degree programs	Science Chemistry, Master of Science Molecular						
realistic asincy of the module to other degree programs	Biotechnology						
	Note: The course(s) not chosen in the compulsory area, (a)						
Other	or (b, c), can additionally be chosen in the elective area.						

Module Handbook	Version	As of	JG
Master of Science Biomedical Chemistry	2.0	14.03.2024	JGIO

Module BCF-P	Practica	al Course in Mol	ecular Biolog	gy and	[Modul-I	(ennnummer ]
	Biochemistry					
Mandatory or elective Module	Elective				-	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Practical Course "Molecular Biology and Biochemistry"	APr	1 (1)	М	9	40,5 h	4,5
b) Supporting seminar to a)	S	1 (1)	М	1	34,5 h	1,5
In order to complete the module, yo	u have to	fulfil the following	requirements:			
Compulsory Attendance	APr					
Active participation	According	g to § 5 para. 3			0	
Coursework						
Module examination				.0		
Qualification Goals, learning outcom	ne, compe	tences				
<ul> <li>apply complex biocher</li> <li>to carry out experiment</li> <li>to work carefully and it</li> <li>interpret the results of</li> <li>apply effective time and</li> <li>to critically question a</li> </ul> Contents <ul> <li>a)</li> <li>Molecular biology: Promodified bacteria.</li> <li>Generation of a bacter</li> <li>Heterologous protein</li> <li>Characterisation of the</li> <li>2D gel electrophoresis</li> <li>physical parameters (if</li> <li>Analysis of the phosphere</li> <li>Staining techniques for</li> <li>Purification of lysozymeric</li> <li>The student elaborate the presentation.</li> <li>The student analyses a</li> </ul>	nts indepenn n a coordi f their expond current to nd resource current to nd scientif oduction of rial strain s expression e protein a scientric protein g norylation protein g ne: ion excl the enzymous s and pres	ndently and on thei nated manner. eriments correctly a e management. opic in biochemistry ically discuss preser in an expression plas suitable for protein of a n expression plas uitable for protein of a cli suitable for protein of a cli suitable for protein of a cli suitable for protein of a stress p point, size) for two- pattern of a stress p rels hange chromatogra e ents a given, curren	r own responsik and document th and defend it in nted seminar lea mid, production expression. in purification. rent stressors (H dimensional sep protein by 2D ge phy, protein pre-	ility using cou hem in an app a a discussion i ctures. a and characte heat, oxidants, baration of con I electrophore ecipitation, SD emistry and fa	ropriate form. in front of the e risation of gene , etc.), use of p mplex protein r esis and Wester S-PAGE, photo ces the audien	entire audience. etically rotein-specific nixtures. 'n blot analysis metric assay to ce to discuss
Compulsory entrance requirements						
Recommended participation require and/or individual courses of the mo		or the module				
Language(s) of instruction and exam	ination		German or Eng	lish		
Weight of the module grade in the o	overall grad	de	Not graded			
Frequency of module offer			Every term in t	he lecture-fre	e period	
Descent for compulsory attendence			A			
Reasons for compulsory attendance			According to H	ochSchG § 26	Para. 2 (7), Pra	ctical Course



Transferability of the module to other degree programs	Bachelor of Science Molecular Biotechnology, Master of Science Chemistry
Other	Only in the lecture-free period

without Buarantee



Module OCF	Organi	Chemistry			[Modul-K	(ennnummer ]
Mandatory or elective Module	м				<u>L</u>	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture "Aromatic/heterocyclic compounds"	L	1 (1)	М	2	69,0 h	3
b) Supporting exercise to a)	Е	1 (1)	м	1	34,5 h	1,5
c) Supporting seminar (Trainee Seminar)	S	1 (1)	М	1	34,5	1,5
In order to complete the module, ye	ou have to	fulfil the following	requirements:			
Compulsory Attendance	S		•			
Active participation		g to § 5 para. 3			U	
Coursework		5 to 3 5 harar 2				
Module examination	Usually w and b)	ritten exam (120 m	in), alternatively	y oral exam (3	0 min) on the c	ontents of a)
Qualification Goals, learning outcor	ne. compe	tences				
<ul> <li>reproduce in-depth kin concepts and method</li> <li>to work out and deep</li> <li>to establish connection chemistry within the sint to transfer the learne</li> <li>to identify problems i work out possible solute valuate them critical</li> <li>present their results i in discussions</li> <li>to critically question at to critically question at to critically question at to analyse the results</li> <li>work independently of develop their preparation analyse the results</li> <li>expand their methodo</li> <li>to work out and plan</li> </ul>	s from the en content ons and link subject and d lecture co n the deve utions inde ly n a compre- and evaluat vidual spec on research tive skills in of indeper ological kno their exper upervisors when deal	se fields and classify s from the field of a s between topics ar with related subject ontents to unknown opment of synthesi pendently by linking hensible manner ar e the solution strate ialisation and perso -related topics in pr ndependently dent literature rese owledge by implement iments and to imple how to carry out th ing with scientific per	w them with rega irromatics and he had contents from ct areas. In tasks is strategies and g the acquired k and using scientif egies developed nal profile build reparative organ earch. enting new apparation ement them ind e experiments a roblems,	ard to their sig eterocyclic che n the field of a l in the answe nowledge with ically correct t l. ing in prepara nic chemistry, aratus and ana ependently, and to correct propriate acti	nificance emistry indeper aromatics and h ring of complex h their own idea erminology and tion for later in alytical procedu them, on,	ndently. neterocyclic questions, to as and to d defend them dependent
<ul> <li>assess the safety aspective of the safety asp</li></ul>	sh languag <sup>.</sup> in a team	and to carry out pre	eliminary work,			experimental
<ul><li>assess the safety aspe</li><li>to develop their Engli</li></ul>	sh languag in a team a team ar theoretica	and to carry out pre d to deal with haza I knowledge throug	eliminary work, rdous substance	es, to analyse a		experimental

Module Handbook
Master of Science Biomedical Chemistry

a)

Version	As of
2.0	14.03.2024



Aromaticity (criteria), systematic treatment of annulenes, non-alternating systems, PAHs, methods of preparation and properties of selected systems, Classification and nomenclatures of heterocycles, physical properties (solubility, pKs, dipole moments, ...) . Systematic treatment of small rings with up to two heteroatoms, medium rings with up to four heteroatoms, sevenand eight-membered rings in their occurrence and production as well as specific reactivity. Application as active substances and in materials science. b) Consolidation of the lecture material and applications in transfer exercises c) Preparative methods, reagents in organic synthesis, reaction types and reaction mechanisms • **Compulsory entrance requirements** Recommended participation requirement(s) for the module and/or individual courses of the module Language(s) of instruction and examination German or English 12/66 Weight of the module grade in the overall grade Frequency of module offer Every term Seminar according to § 5 para. 5: The learning objectives are based on direct interaction between students. In Reasons for compulsory attendance addition to practical professional competence, important learning objectives are literature research, presentation and leading discussions. Person responsible for the module N.N. • Master of Science Chemistry Transferability of the module to other degree programs Recommended literature: Gilchrist: Heterocyclenchemie, Other hormation ۰. Joule/Mills: Heterocyclic Chemistry, Brückner: Reaktionsmechanismen



Module OCF-P	Practica	al Course on Mo	lecule Synth	esis	[Modul-k	Kennnummer ]
Mandatory or elective Module	м				<u>_</u>	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)         Mandatory/ elective         Contact Time (SWH)         Self Study				
Practical Course "Molecule Synthesis"	APr	2 (2)	М	12	54 h	6
In order to complete the module, yo	u have to	fulfil the following	requirements:			
Compulsory Attendance	APr					0
Active participation	According	to § 5 para. 3				
Coursework						
Module examination						
Qualification Goals, learning outcom	e, compet	ences				
<ul> <li>develop their preparat</li> <li>analyse the results of i</li> <li>extend their methodol</li> <li>work out and plan their</li> <li>debate with their supe</li> <li>work out solutions who</li> <li>assess the safety aspect</li> <li>develop their English la</li> <li>plan tasks together in a</li> <li>work responsibly in a t</li> <li>analyse and correct the</li> <li>assess and optimise th</li> </ul> Contents Preparation of 3-4 research-related p from current chemical journals or Orget	ndepende ogical kno ir experime ervisors ho en dealing cts of chen anguage sl a team and eam and t e experime e results o	nt literature researd wledge by impleme ents and to impleme w to carry out the e with scientific prob nicals and experime cills through English d to carry out prelin o handle hazardous ental results on the f the experiments.	enting new appa ent them indepe- experiments and olems, ints and take ap -language litera hinary work, s substances, basis of theoret	endently, I to correct the propriate acti- ture and supe tical knowledg	em, on, ervisors, ge through tech	nical literature,
Compulsory entrance requirements	Same Synth					
Recommended participation require and/or individual courses of the mod	dule	or the module	Cormon or Fra	lich		
Language(s) of instruction and exam			German or English			
Weight of the module grade in the o	veran grad		Not graded			
Frequency of module offer Reasons for compulsory attendance			Every term According to HochSchG § 26 Para. 2 (7), Practical Course			
Person responsible for the module			UnivProf. Dr. Till Opatz			
Transferability of the module to othe	er degree	programs	Master of Scier	•		
Other			Recommended • Gilch • Joule • Brüch	I Literature: rist: Heterocyo /Mills: Hetero <ner: reaktion<br="">nic Synthesis,</ner:>		1

Module PMC2	Pharma	Pharmacology for Natural Scientists			[Modul-I	(ennnummer ]	
Mandatory or elective Module	м	И					
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	ter					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoint	
a) Lecture "Pharmacology for Natural Scientists"	L	1 (1)	М	3	103,5 h	4,5	
b) Supporting seminar to a)	S	1 (1)	М	1	34,5 h	1,5	
In order to complete the module, yo	ou have to	fulfil the following	requirements:				
Compulsory Attendance							
Active participation	According	g to § 5 para. 3					
Coursework					0		
Module examination	Usually w and b)	rritten exam (60 mir	n), alternatively	oral exam (15	min) on the co	ntents of a)	
Qualification Goals, learning outcon	ne, compe	tences					
and molecular biology to name important dr adverse effects and in Contents a)	ugs for the teractions		non diseases an	d their most ir	nportant prope	erties (includii	
<ul> <li>Principles of pharmacodyn</li> <li>Principles of pharmacokine</li> <li>Principles of pharmacogen</li> <li>Drug-drug interactions</li> <li>Important transmitters</li> <li>Drugs for the treatment or cardiovascular diseases, diable</li> <li>processes of drug research</li> <li>Job profiles for natural scie</li> </ul>	etics etics preventio abetes, me	ental diseases, tumo	ur diseases) n,	autoimmune d	iseases, infecti	ons,	
Compulsory entrance requirements			,				
Recommended participation require and/or individual courses of the mo	ement(s) fo	or the module					
Language(s) of instruction and exam			German or Eng	lish			
Weight of the module grade in the o		de	6/66				
Frequency of module offer	0		Every term				
Reasons for compulsory attendance			,				
Person responsible for the module			apl. Prof. Dr. El	len Closs			
Transferability of the module to oth	er degree	programs					
Other			b) Can also be an online semi		he form of a bl	ock seminar	



Module PMC3	Pharma	aceutical Science	es for Natura	l Scientists	[Modul-	Kennnummer ]	
Mandatory or elective Module	м						
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ Contact elective Time (SWH) Self Study Credit				
a) Lecture "Principles and Special Aspects of Drug Design"	L	2 (2)	м	2	69 h	3	
b) Supporting seminar to a)	S	2 (2)	М	2	69 h	3	
In order to complete the module, y	ou have to	fulfil the following	requirements:				
Compulsory Attendance							
Active participation	According	g to § 5 para. 3					
Coursework					2		
Module examination	Usually w and b)	ritten exam (120 m	in), alternativel	y oral exam (30	) min) on the c	contents of a)	
Qualification Goals, learning outco	me, compe	tences					
<ul> <li>to work out and press</li> <li>discuss medicinal che</li> <li>to assess and classify</li> </ul> Contents <ul> <li>a) and b)</li> <li>Principles of drug dest</li> <li>Drug targets</li> <li>Hit-to-lead developm</li> <li>Optimisation of lead</li> <li>Concepts of structure</li> <li>Analysis of target-liga</li> <li>ADME-tox properties</li> <li>PK-PD optimisation of</li> <li>Development of cova</li> <li>Determination and ca</li> <li>Drug monitoring</li> <li>Literature / publication</li> <li>Computer-aided drug pharmacophore mod structure optimisation</li> </ul>	emistry topi new appro sign nent structures e-activity re and interact of drugs; ru f drugs (pha alculation o ons on curro g design (CA lelling, protein n, generatio	cs appropriately aches to drug devel lationships and SAR ions at molecular a elationship with che armacodynamics-ph f physicochemical p ent topics in drug de DD): e.g. visualisati ein-ligand docking, v on of 3D structures	opment and op analysis nd atomic level emical structure narmacokinetics arameters of ac evelopment on and analysis virtual screening of small molecu	timisation ) tive substance of protein-liga g, scoring, QSA les, force field	nd complexes R, ADME mod s, MD simulati	elling, lead ons,	
conformational analy design, target assess Biologics and antiboo Nucleic acids as drug Drug delivery, drug ta Combinatorial appro Compulsory entrance requirement Recommended participation requirement	ment lies s and drug t argeting, dr aches s rement(s) fo	argets ug transport			- <sup>1</sup> 2 11 - 11 - 12 - 12 - 12 - 12 - 12 -	מ מכזוצוו, ווטו מו י	
and/or individual courses of the m							
	mination		German or Eng				

Weight of the module grade in the overall grade	6/66
Frequency of module offer	Every term
Reasons for compulsory attendance	
Person responsible for the module	UnivProf. Dr. Tanja Schirmeister
Transferability of the module to other degree programs	
Other	
hormation	hout evante



Module FMP	Research Project				[Modul-ł	(ennnummer ]	
Mandatory or elective Module	м				<u>+</u>		
Creditpoints (LP) and workload	12 LP = 3	60 h					
Module duration (according to course plan)	1 Semest	Semester					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Practical Course "Research Paper"	Apr	3 (3)	М	22	99,0 h	11	
b) Supporting Seminar "Guidance for independent scientific work"	S	3 (3)	М	1	19,5 h	C1	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance	Apr (acco	rding to the task an	d agreement wi	th the supervi	sor)		
Active participation	According	g to § 5 para. 3, pres	entation on the	e research mod	dule (30 min)		
Coursework	Maintena	nce of a laboratory	notebook		0		
Module examination	Research	report					
Qualification Goals, learning outcom	e, compet	tences					
recorded reproducibly in a laboratory By working in a working group, the st <b>Contents</b> a) Participation in a current research b) Introduction to planning, executior in a report (protocol) and seminar lec	udents ex project in n and docu	pand their commun the chosen working umentation of more	ication and team	mwork skills. I in the study p	programme.		
Compulsory entrance requirements	luie.		According to P	0			
Recommended participation require and/or individual courses of the moc		or the module		-			
Language(s) of instruction and exami	ination		German or English				
Weight of the module grade in the o	verall grad	de	Not graded				
Frequency of module offer			Every term				
Reasons for compulsory attendance			According to HochSchG § 26 Abs. 2 (7), scientific (practical research work/internship (according to assignment and agreement with the supervisor).				
Person responsible for the module			All full-time lec	turers involve	d in the degree	e programme	
Transferability of the module to othe	er degree	programs	Master of Scier	nce Biomedicir	nal Chemistry		
Other			The module tal Individual supe in a participati to semester. Ex application.	ervision; the nung working gro	umber of interr oup may vary fr	om semester	

Version	As of	JG
2.0	14.03.2024	JG



Module MSC	Master Thesis				[Modul	-Kennnummer ]	
Mandatory or elective Module	м		<u>k</u>				
Creditpoints (LP) and workload	30 LP = 9	00 h					
Module duration (according to course plan)	1 Semest	1 Semester					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
Master Thesis		4 (4)	М	6 months all day	900 h	30	
In order to complete the module, yo	u have to	fulfil the following	requirements:		ł		
Compulsory Attendance	Master th	esis (according to a	ssignment and a	agreement wi	th the supervi	isor)	
Active participation	According	g to § 5 Para. 3, pres	entation on the	Master's the	sis (30 min)		
Coursework	Maintena	nce of a laboratory	notebook				
Module examination	Master th	esis			0		
Qualification Goals, learning outcom	ne, compet	tences					
Contents Master's thesis: Composition of a scie introduction including objectives, ma to document further primary data. Pi	terial & m	ethods as well as re	sults, discussior	n, bibliography	; an appendi		
Compulsory entrance requirements			According to §				
Recommended participation require and/or individual courses of the mo		or the module					
Language(s) of instruction and exam	ination		German or English				
Weight of the module grade in the o	verall grad	de	30/66				
Frequency of module offer			Every term				
Reasons for compulsory attendance			According to HochSchG § 26 Para. 2 (7), scientific (practical) research work/internship (according to assignment and agreement with the supervisor).				
Person responsible for the module			All full-time lecturers involved in the degree programme				
Transferability of the module to other degree programs			Master of Science Biomedicinal Chemistry				
Other			The module takes place full-time for 6 months. Individual supervision; the number of internships offered in a participating working group may vary from semester to semester. External Master's thesis possible on application.				
11.							

Module Handbook	Version	As of	JG
Master of Science Biomedical Chemistry	2.0	14.03.2024	JGIO

### **Elective Area**

All elective modules do not count towards the final grade.

Module NC	Chemistry of Natural Products [Modul-Kennnu					(ennnummer ]	
Mandatory or elective Module	Elective	Elective					
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lecture "Chemistry of Natural Products"	L	2 (1)	М	2	69 h	3	
b) Supporting exercise to a)	E	2 (1)	М	1	34,5 h	1,5	
c) Seminar "Retrosynthesis"	S	2 (1)	М	1	34,5 h	1,5	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance							
Active participation	According	to § 5 para. 3					
Coursework				<u></u>			
Module examination	Usually w b) and c)	ritten exam (120 m	in), alternativel	y oral exam (3	0 min) on the c	ontents of a),	
Qualification Goals, learning outcom	e, compet	ences					
<ul> <li>to independently work</li> <li>to establish connection within the subject and</li> <li>to transfer the content</li> <li>to identify problems in independently work ou evaluate them</li> </ul>	ns and link with relat s of the le the devel at possible	s between topics ar ed subject areas. cture to unknown t opment of synthesi solutions by linking	nd contents from asks. s strategies and g the acquired k	m the field of r I in the answer nowledge wit	natural product ring of complex	questions, to	
to critically question ar Contents	nd evaluat	e the solution strate	egies developed	1.			
<ul> <li>a)</li> <li>Organic Chemistry 5: Classe</li> <li>Amino acids, peptides and p</li> <li>Terpenes and steroids</li> <li>Lipids and eicosanoids</li> <li>Polyketides</li> <li>carbohydrates</li> <li>Biogenic amines and alkaloi</li> <li>nitrogenous cofactors of prosents and biosynthesis and</li> <li>b)</li> <li>consolidation of the lecture mate</li> </ul>	oroteins, p oteins and analys erial and a	is of natural produc	ts. fer exercises	es and nucleic	: acids, nucleic a	acid synthesis.	
<ul> <li>Methods of organic synthesis an</li> </ul>	d retrosyr	thesis on concrete	examples.				
Compulsory entrance requirements							
Recommended participation require and/or individual courses of the mod		or the module					

Language(s) of instruction and examination	German or English
Weight of the module grade in the overall grade	Not graded
Frequency of module offer	Only in the summer term
Reasons for compulsory attendance	
Person responsible for the module	UnivProf. Dr. Till Opatz
Transferability of the module to other degree programs	Master of Science Chemistry
Other	Recommended Literature: Nuhn: Naturstoffchemie Habermehl/Hammann/Krebs/Ternes: Naturstoffchemie
hormation	outebarante



Module RPC	Radiop	harmaceutical C	[Modul-K	ennnummer ]				
Mandatory or elective Module	Elective	lective						
Creditpoints (LP) and workload	6 LP = 18	0 h						
Module duration	2 Semest	or						
(according to course plan)	2 Jennest	- Seniester						
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints		
a) Lecture "Radiopharmaceutical Chemistry 1"	L	1 (2)	М	2	69h	3		
b) Lecture "Radiopharmaceutical Chemistry 2"	L	2 (1)	М	2	69 h	3		
In order to complete the module, ye	ou have to	fulfil the following	requirements:					
Compulsory Attendance								
Active participation	According	g to § 5 para. 3			$\sim$			
Coursework								
Module examination	Usually or and b)	ral exam (30 min), a	lternatively wri	tten exam (12	0 min) on the co	ontents of a)		
Qualification Goals, learning outcor	ne, compe	tences						
<ul> <li>preclinical a</li> <li>radionuclida</li> <li>radiopharm</li> <li>properties,</li> <li>RPC in onco</li> </ul>	Chemistry n and basic nd clinical e productio aceutical p production logy, neuro	(RPC) are offered a s of RPC: decay mod imaging techniques, n in RPC: cyclotron, rocedures in diagno , labelling chemistry logy and other field	of new radiopha s block courses les, shielding & reactor & gene stics and therap v & application ls of application	over 2 semesi detection. rator, by: SPECT, PET of relevant nu	ters. Contents a	re:		
This module builds on the basic know		he lecture "Introduc	tion to Nuclear	Chemistry".				
Compulsory entrance requirements Recommended participation require and/or individual courses of the mo	ement(s) fo	or the module	Module "Introduction to Nuclear Chemistry"					
Language(s) of instruction and exan			German or English					
Weight of the module grade in the overall grade			Not graded					
Frequency of module offer			a) Only in the winter term b) Only in the summer term					
Reasons for compulsory attendance	9							
Person responsible for the module			UnivProf. Patrick Riß					
Transferability of the module to oth	er degree	programs	Master of Science Chemistry					
Other								

	<b>-</b>			· · · · · ·	[Diadul k	(		
Module MCP	Practical Course Selected Aspects of Medicinal [Modul-Kennnum]					tennnummer j		
	-	Chemistry						
Mandatory or elective Module	Elective							
Creditpoints (LP) and workload	6 LP = 18	0 h						
Module duration (according to course plan)	1 Semest	er						
		Regular term						
Courses/ Learning formats	Туре	e Regular term when starting in Winter term (Summer term) Handatory/ (Summer term) Contact Time (SWH) Self Study Creditp						
a) Practical Course "Selected Aspects of Medicinal Chemistry"	APr	1 - 3 (1 - 3)	М	6	117 h	6		
In order to complete the module, yo	u have to	fulfil the following	requirements:					
Compulsory Attendance	APr							
Active participation	According	g to § 5 para. 3						
Coursework					5			
Module examination								
Qualification Goals, learning outcom	e. compe	ences						
chemistry and pharma use the subject-specifi point out connections optimisation of active s to work out and preser discuss medicinal chen to evaluate and classify to understand and apply understand and apply understand and apply understand and apply prepare and analyse ey discuss, summarise and	c terminol and differ substance nt a medic nistry topi y new app ly methods experiment methods f methods f	ences between diffe s inal chemistry topic cs appropriately roaches to drug dev ls of drug and drug ntal and theoretical or biotransformatic or determining the or computer-aided al data	erent approache independently velopment and analysis methods for the on of active subs efficacy of biolo drug design.	optimisation e determination stances and fo ogically active	on of physico-cl r drug monitori substances.	hemical drug		
Contents								
The practical course includes the folic Visulisation and analys Calculation of physicod Pharmacophore mode Protein-ligand docking Homology modelling Lead structure optimis Biotransformation and Stability studies of drug Drug and drug substan Determination of phys Drug monitoring Enzyme kinetics, ligand Quantitative HPLC for o	is of prote hemical a ls ation determin gs ce analysi icochemic	in-ligand complexed nd pharmacokinetic ation of metabolite s al and pharmacokin tudies	s parameters	S				
Compulsory entrance requirements								
Recommended participation require and/or individual courses of the mod		or the module						
Language(s) of instruction and exam	ination		German or Eng	lish				
Weight of the module grade in the o	verall grad	de	Not graded					

Module Handbook	Version	As of	JG
Master of Science Biomedical Chemistry	2.0	14.03.2024	JGO

Frequency of module offer	Every term in the lecture-free period
Reasons for compulsory attendance	According to HochSchG § 26 Para. 2 (7), Practical Course
Person responsible for the module	UnivProf. Dr. Tanja Schirmeister
Transferability of the module to other degree programs	
Other	Block internship (3 weeks full time) during the lecture-free period
hormation	outebarante

Module BAC	Bioinorganic Chemistry					Kennnummer ]		
Mandatory or elective Module	Elective	Elective						
Creditpoints (LP) and workload	6 LP = 18	0 h						
Module duration (according to course plan)	1 Semest	er						
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints		
a) Lecture "Bioinorganic Chemistry"	L	2 (1 o. 3)	М	3	103 <i>,</i> 5 h	4,5		
b) Supporting Seminar to a)	S	2 (1 o. 3)	М	1	34,5 h	1,5		
In order to complete the module, yo	u have to	fulfil the following	requirements:					
Compulsory Attendance	S							
Active participation	According	g to § 5 para. 3						
Coursework				5	0			
Module examination	Usually w	ritten exam (120 m	in), alternatively	y oral exam (3	0 min) on the c	contents of a)		
Qualification Goals, learning outcom	ie, compet	tences						
<ul> <li>can transfer the learner</li> <li>can establish connectidisciplines,</li> <li>have gained an unders</li> </ul> Contents Bioinorganic chemistry is a cross-sect The lecture serves to identify the spe Biological processes such as photosyn Selected examples of metalloprotein	ons and lir standing of cional disci cific roles nthesis or	iks between topics a f the significance of pline of biochemisti of certain metal ion cellular respiration a	metal ions in liv metal ions in liv ry and coordina s in chemical-bi are discussed.	tion chemistry	r. Icesses.			
in more detail as well as electron tran						-,		
Compulsory entrance requirements								
Recommended participation require and/or individual courses of the model of the mod		or the module						
Language(s) of instruction and exam	ination		German or English					
Weight of the module grade in the o	verall grad	de	Not graded					
Frequency of module offer			Only in the sun	nmer term				
Reasons for compulsory attendance			Upper seminar according to § 5 para. 5: The learning objectives are based on direct interaction between students. In addition to practical professional competence important learning objectives are literature research, presentation and leading discussions.					
Person responsible for the module			UnivProf. Dr. Eva Rentschler					
Transferability of the module to oth	er degree	programs	Master of Scier	nce Chemistry				
Other								



Module BPC	Biophy	sical Chemistry			[Modul-I	Kennnummer ]		
Mandatory or elective Module	Elective	Elective						
Creditpoints (LP) and workload	6 LP = 18	0 h						
Module duration (according to course plan)	1 Semest	1 Semester						
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints		
a) Lecture "Biophysical Chemistry"	L	2 (1)	М	2	69 h	3		
b) Supporting exercise to a)	Е	2 (1)	М	2	69 h	3		
In order to complete the module, yo	ou have to	fulfil the following	requirements:					
Compulsory Attendance					X	0		
Active participation	According	g to § 5 para. 3			5			
Coursework								
Module examination	Usually w	ritten exam (120 m	in), alternatively	y oral exam (3	0 min)			
Qualification Goals, learning outcon	ne, compe	ences						
get to the bottom of unknown pheno Contents a) Basics of modern biophysical meth Membrane transport, phase transitic motors, single molecule techniques, the drug discovery process. b) In-depth or supplementary topics	nods with e ons in mem Raman sca	branes, nanopartic ttering, thermodyn	le sensors, rate amics of chemic	equations and cal bonds, phy	l dynamics in c sical-chemical			
Compulsory entrance requirements				•	•			
Recommended participation require and/or individual courses of the mo		or the module						
Language(s) of instruction and exam	nination		German or Eng	lish				
Weight of the module grade in the c	overall grad	de	Not graded					
Frequency of module offer			Only in the sun	nmer term				
Reasons for compulsory attendance								
Person responsible for the module			UnivProf. Dr.	Carsten Sönni	chsen			
Transferability of the module to other degree programs			Master of Scier	nce Chemistry				
Other								
			1					

Module Tox1	Toxicol	ogy 1			[Modul-H	(ennnummer ]
Mandatory or elective Module	Elective					
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration						
(according to course plan)	1 Semest	er				_
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture "General Toxicology"	L	1 - 3 (1 - 3)	М	2	69 h	3
b) Seminar "Molecular and Cellular Toxicology"	S	1 - 3 (1 - 3)	М	2	69 h	3
In order to complete the module, yo	ou have to	fulfil the following	requirements:			
Compulsory Attendance	S					
Active participation	According	g to § 5 para. 3				
Coursework					0	
Module examination	Usually w	ritten exam (60 mir	n), alternatively	oral exam (15	min) on the co	ntents of a)
Qualification Goals, learning outcom	ne, compe	tences				
<ul> <li>name the treated biog</li> <li>describe the symptom</li> <li>apply their knowledge toxin and (theoreticall</li> <li>apply their knowledge fine dust, carcinogenic</li> <li>b) The students are able to         <ul> <li>independently process original publications o</li> </ul> </li> <li>Contents         <ul> <li>a)</li> <li>Basics of toxicology</li> <li>Toxic effects (receptor</li> <li>Chemical mutagenesis</li> <li>Heavy metals, solvent:</li> <li>Biological toxins (plant</li> <li>Biocides, toxic drugs, to</li> </ul> </li> </ul>	atology of to conclud y) to carry to be able substance s original p rally, evalut site theor s and carcin s, alcohols, t, bacterial	the different poison de on the basis of th out an appropriate to assess current to es, new psychoactiv ublications from the nate them and discu	nings. ne symptomatol therapy. opics in the med e substances wh e field of molect ss them criticall concentration to ns Il toxins)	ogy a specific dia. This incluc nich are releva ular toxicology y within the g	poisoning and t des topics such ant to the public /, present the d roup.	the triggering as exposure to c.
<ul> <li>therapy of poisoning</li> <li>b) In the seminar, the topics of chem the field of genotoxicology are discus cytotoxicity, mechanisms of carcinog</li> </ul>	ssed. This i	ncludes topics such	as ageing resea	rch (senescen		
Compulsory entrance requirements						
Recommended participation require and/or individual courses of the mo		or the module				
Language(s) of instruction and exam			German			
Weight of the module grade in the o	overall grad	de	Not graded			
Frequency of module offer			Every term			
Reasons for compulsory attendance			Seminar according to § 5 para. 5: The learning objectives are based on direct interaction between students. In addition to practical professional competence, important learning objectives are literature research, presentation and leading discussions.			dents. In ce, important
Person responsible for the module			UnivProf. Dr.	Markus Christ	mann	

Module Handbook	Version	As of	JGU
Master of Science Biomedical Chemistry	2.0	14.03.2024	19/0

Transferability of the module to other degree programs	Bachelor of Science Biomedical Chemistry
Other	

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Module ToxP	Toxicol	ogy 2			[Modul-I	Kennnummer ]	
Mandatory or elective Module	Elective						
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	ng in Mandatory/ Contact erm elective Time (SWH) Self Study C				
Practical Course "Molecular methods in toxicology"	APr	2 (1 o. 3)	М	6	117 h	6	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance	APr					0	
Active participation	According	g to § 5 para. 3					
Coursework							
Module examination	Written e	xam (30min)			0		
Qualification Goals, learning outcom	e, compe	tences					
<ul> <li>name toxic effects of ratio carry out toxicologic analyses, microscopic et to adequately record a</li> <li>Contents</li> <li>In the context of the practical course, mutagenesis, genotoxicity, DNA repairelevant techniques.</li> <li>Mechanisms of cytotox</li> <li>Investigation of genoto</li> <li>Transcriptional and epi</li> <li>Principles of toxicologi</li> <li>molecular causes of ag</li> <li>post-translational modified</li> </ul>	ally releva examination nd evalua students ir and cell kicity (apo oxic effect city assays igenetic re cal risk ass eing	ant examinations (d ons). te the performance should acquire furt death mechanisms, ptosis, necrosis, aut s: SCE, aberration, p , Ames test egulatory mechanism sessment	etermination of and results of p her theoretical l as well as pract cophagy,) point mutation a	cytotoxicity a practical invest knowledge abo tical knowledg	nd genotoxicit <sup>,</sup> tigations. out mechanism	y, expression	
Compulsory entrance requirements			Module Toxico	logy 1			
Recommended participation require and/or individual courses of the modes and the modes of the modes and the modes are as a set of the modes are a		or the module					
Language(s) of instruction and examination			German				
Weight of the module grade in the overall grade			Not graded				
Frequency of module offer			Only in the summer term				
Reasons for compulsory attendance			According to HochSchG § 26 Para. 2 (7), Practical Course				
Person responsible for the module UnivProf. Dr. Markus Christmann							
Transferability of the module to othe	er degree	programs	Bachelor of Sci	ence Biomedio	cal Chemistry		
Other							

Module Immun1	<mark>Immun</mark>	ological Principl	es		[Modul-K	ennnummer ]
Mandatory or elective Module	Elective				<u>_</u>	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture "Immunological Principles"	L	2 (1 o. 3)	М	2	69 h	3
b) Supporting seminar to a)	S	2 (1 o. 3)	М	2	69 h	3
In order to complete the module, yo	u have to	fulfil the following	requirements:			
Compulsory Attendance						
Active participation	According	g to § 5 para. 3				
Coursework					0	
Module examination	Usually w and b)	ritten exam (90 mir	n), alternatively	oral exam (30	min) on the co	ntents of a)
Qualification Goals, learning outcom	e, compe	tences				
<ul> <li>establish the significan</li> <li>understand the special allergies and autoimm</li> <li>independently develop</li> <li>discuss immunological</li> </ul> Contents Lecture and seminar include the follo <ul> <li>Organs and cells of the</li> <li>Mechanisms of innate</li> <li>Development and funct</li> <li>Development and funct</li> <li>Tolerance mechanisms</li> <li>Importance of the maj</li> <li>Genetic models in imm</li> <li>Signal transduction in l</li> <li>Mucosal immune syste</li> <li>Mechanisms of infection</li> </ul>	I significan une diseas o and pres topics app wing topic immune immunity ction of B-o ction of T c or histoco nunology lymphocytem	ce of immunologica ees. ent a (given) immur propriately. cs: system; Haematopo cells and antibodies eells mpatibility complex es	al research for the nological topic.	ne developme		
Compulsory entrance requirements						
Recommended participation require and/or individual courses of the mod		or the module				
Language(s) of instruction and exam			German			
Weight of the module grade in the o	rade in the overall grade Not graded					
Frequency of module offer	Only in the summer term					
Reasons for compulsory attendance						
Person responsible for the module			apl. Prof. Dr. N	lichael Stasser	1	
Transferability of the module to othe	er degree	programs			cal Chemistry, N cience Biomedic	
Other					st half of the se he second half	

Module Immun2	<b>Practica</b>	al Exercises in In	nmunology		[Modul-k	(ennnummer ]	
Mandatory or elective Module	Elective						
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Practical Exercise "Immunology"	E	2 (1 o. 3)	М	8	96 h	6	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance	E					20	
Active participation	According	to § 5 para. 3			X	0	
Coursework							
Module examination							
Qualification Goals, learning outcom	e, compet	ences					
<ul> <li>document the results of statistical methods</li> <li>agree on individual wo</li> <li>reproduce and explain</li> </ul> Contents In the exercises, the following content <ul> <li>Quantification of cytok</li> <li>Detection of mediator</li> <li>Determination of the a</li> <li>Identification and enrice</li> <li>Blood group serology</li> <li>Enrichment and activation</li> </ul>	rk steps, p the theor ts will be tines by EL release fro ctivity of r chment of	lan them together a y on which the expe worked on experime ISA and qRT-PCR om activated mast o reporter genes defined cell popula	and implement rriments are bas entally: cells tions using FAC	them in a coo sed	rdinated mann		
Compulsory entrance requirements			Module "Immu	inological Prin	ciples"		
Recommended participation require and/or individual courses of the mode and the mode state and the mode state and the state an		or the module					
Language(s) of instruction and examined and examin			German				
Weight of the module grade in the o	verall grad	le	Not graded				
Frequency of module offer			Only in the sun	nmer term du	ring the lecture	-free period	
Reasons for compulsory attendance			According to H	ochSchG § 26	Abs. 2 (7), prac	tical exercise	
Person responsible for the module			apl. Pof. Dr. Mi	ichael Stassen			
Transferability of the module to other degree programs				Bachelor of Science Biomedical Chemistry, Master of Science Biology, Master of Science Biomedicine			
Other			Registration re course during t	•		er; block	

Module Handbook	Version	As of	JG
Master of Science Biomedical Chemistry	2.0	14.03.2024	JGIO

Module PB1	Pharma	Pharmaceutical Biology				[Modul-Kennnummer ]	
Mandatory or elective Module	Elective	Elective					
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lecture "Pharmaceutical Biology I, II or III"	L	1 - 3 (1 - 3)	М	2	69 h	3	
b) Seminar "Biogenic Medicinal Products (Antibiotics, Genetically Engineered Medicinal Products)"	s	1 - 3 (1 - 3)	М	2	69 h	3	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance	S			5	0		
Active participation	According	g to § 5 para. 3					
Coursework				10			
Module examination	Usually w and b)	ritten exam (120 m	in), alternativel	y oral exam (30	0 min) on the c	ontents of a)	
Qualification Goals, learning outcom	e, compe	tences					
a) and b) The students are able to, • classify and reproduce	basic theo	oretical knowledge	of pharmaceutio	cal biology			
Contents							
<ul> <li>a) Medicinal plants, biogenic and nor</li> <li>b) Antibiotics, plant cytostatics, gene</li> <li>pharmaceutical biology</li> </ul>					abolites, techn	ical methods of	
Compulsory entrance requirements							
Recommended participation require and/or individual courses of the mod		or the module					
Language(s) of instruction and exam	ination		German				
Weight of the module grade in the o	verall grad	de	Not graded				
Frequency of module offer			a) Every semester (I, II and III alternately) b) Every semester				
Reasons for compulsory attendance			Seminar according to § 5 para. 5: The learning objectives are based on direct interaction between students. In addition to practical professional competence, important learning objectives are literature research, presentation and leading discussions.				
Person responsible for the module			UnivProf. Dr. Thomas Efferth				
Transferability of the module to othe	er degree	programs	Bachelor of Sci	ence Biomedic	cal Chemistry		
Other							

Module Handbook	Version	As of	JG
Master of Science Biomedical Chemistry	2.0	14.03.2024	1910

Module PBP	Practic	al Course in Pha	Pharamaceutical Biology [Modul-Kennnummer						
Mandatory or elective Module	Elective								
Creditpoints (LP) and workload	6 LP = 18	0 h							
Module duration (according to course plan)	1 Semes	ter	-				-		
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self St	tudy	Creditpoints		
a) Practical course "Pharmaceutical Biology III: Biological and Phytochemical Investigations".	APr	1 - 3 (1 - 3)	М	6	117 h		6		
In order to complete the module, yo	ou have to	fulfil the following	requirements:						
Compulsory Attendance	APr					$\sim$			
Active participation	According	g to § 5 para. 3							
Coursework									
Module examination									
Qualification Goals, learning outcon	ne, compe	tences							
<ul> <li>separate plant drug m apply basic molecular</li> <li>Contents</li> <li>Biological and phytochemical studies</li> <li>MS, HPLC, isolation of genomic DNA,</li> </ul>	biological s of medici	techniques. nal plants, identifica		< <u>&gt;</u>					
Compulsory entrance requirements			Module "Pharr	naceutical Bio	logy"				
Recommended participation require and/or individual courses of the mo		or the module							
Language(s) of instruction and exam	nination		German						
Weight of the module grade in the o	overall gra	de	Not graded						
Frequency of module offer			Every term						
Reasons for compulsory attendance	9		According to HochSchG § 26 Para. 2 (7), Practical Course						
Person responsible for the module			UnivProf. Dr. Thomas Efferth						
Transferability of the module to oth	er degree	programs	Bachelor of Science Biomedical Chemistry						
Other									
Inform									

Module MiBiT	Microbiology and Biotechnology [Modul-Ken					(ennnummer ]				
Mandatory or elective Module	Elective				<u> </u>					
Creditpoints (LP) and workload	6 LP = 18	0 h								
Module duration (according to course plan)	2 Semest	ters								
Courses/ Learning formats	Type (Summer term) Type (Summer term)			Contact Time (SWH)		Selt Study Creditnoir				
a) Lecture "Microbiology"	L	1 o. 3 (2)	М	2	69 h	3				
b) Lecture "Biotechnology"	L	1 o. 3 (2)	М	2	69 h 3					
In order to complete the module,	you have to	fulfil the following	requirements:							
Compulsory Attendance					X	0				
Active participation	According	g to § 5 para. 3			5					
Coursework										
Module examination	Both exar	le, lecture (25 min), minations must be p xaminations.				ithmetic mean				
Qualification Goals, learning outco	me, compe	tences								
<ul> <li>to name the most im</li> <li>to evaluate the impose</li> <li>b) The students are able to</li> <li>apply in-depth known fermentation, process</li> <li>Interpret biotechnolic</li> <li>extract scientific dat</li> <li>to plan sophisticated</li> <li>to confidently assess</li> </ul>	ledge in imp ssing of prot ogical facts. a from datal l biochemica	acteria in nature and portant sub-areas of ceins and secondary bases al and biotechnologi	d for humans biotechnology metabolites fro ical experiments	(isolation and m submerged	cultures of fun					
Contents										
<ul> <li>a) Microbiology:         <ul> <li>Structure of a bacter</li> <li>Identification and cu</li> <li>Detection of mutation</li> <li>Regulation in bacteri</li> <li>Biotechnology: Theory on the</li> <li>handling of microorg</li> <li>media optimisation f</li> </ul> </li> </ul>	Iture techni ons; metabo ia; structure ganisms and for fermenta	ques of bacteria lic physiology of bac and properties of b fermentation of mi ations of microorgar	acteriophages croorganisms,							
<ul><li>Isolation of biologica</li><li>Isolation of enzymes</li></ul>	from cultur									
<ul> <li>Isolation of biologica</li> <li>Isolation of enzymes</li> <li>Characterisation of a</li> </ul>	from cultur active ingred		[							
Isolation of biologica     Isolation of enzymes     Characterisation of a	from cultur active ingred <b>s</b>	lients.								
Isolation of biologica     Isolation of enzymes     Characterisation of a Compulsory entrance requirement Recommended participation requi	from cultur active ingred s rement(s) for	lients.								
<ul><li>Isolation of biologica</li><li>Isolation of enzymes</li></ul>	from cultur active ingred is rement(s) fo odule	lients.	German							
Isolation of biologica Isolation of enzymes Characterisation of a Compulsory entrance requirement Recommended participation requi and/or individual courses of the m	from cultur active ingred ss rement(s) for odule mination	or the module	German Not graded							
<ul> <li>Isolation of biologica</li> <li>Isolation of enzymes</li> <li>Characterisation of a</li> <li>Compulsory entrance requirement</li> <li>Recommended participation requi</li> <li>and/or individual courses of the m</li> <li>Language(s) of instruction and exa</li> </ul>	from cultur active ingred ss rement(s) for odule mination	or the module								

Person responsible for the module	UnivProf. Dr. Ralf Heermann
Transferability of the module to other degree programs	Bachelor of Science Biomedical Chemistry
Other	

unformation without Buarantee

Module TPhys	<mark>Animal</mark>	<b>Physiology</b>	[Modul-Kennnummer ]				
Mandatory or elective Module	Elective				<u> </u>		
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	) Self Study Creditpo		
a) Lecture "Physiology, Neurobiology and Behaviour of Animals"	L	1 - 3 (1 - 3)	М	4	138 h 6		
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance						6	
Active participation	According	g to § 5 para. 3					
Coursework							
Module examination	Usually w	ritten exam (60 mir	), alternatively	oral exam (30	min)		
Qualification Goals, learning outcom	e, compet	tences		A.4			
<ul> <li>to transfer exemplary a to express themselves</li> <li>Contents</li> <li>Function and interaction ecophysiological adapt</li> <li>Regulation of homeost</li> <li>Biochemistry of enzym</li> <li>Function and mode of</li> <li>Cellular excitability, exists</li> <li>Sensory physiology (e.a)</li> <li>Neurophysiology, learr</li> <li>Behavioural physiology</li> <li>Processes in muscle complexity</li> </ul>	competer on of organ ations to o asis es action of h citation pr g. sight, he ning and m r, orientat	ns extreme habitats normones ocesses, neuronal p earing, sense of bala nemory ion services, interna	f animal experin processing mech nnce, taste, sme al clock	ments. nanisms II)			
Performance physiolog	SY						
Compulsory entrance requirements Recommended participation require and/or individual courses of the mod		or the module					
Language(s) of instruction and exam	ination		German				
Weight of the module grade in the o	verall grad	de	Not graded				
Frequency of module offer			Every term				
Reasons for compulsory attendance							
Person responsible for the module			UnivProf. Dr.	Roland Strauß			
Transferability of the module to othe	er degree	programs	Bachelor of Sci Science Molecu		cal Chemistry, B llogy	Bachelor of	
Other							

Module PPhys	Plant Physiology					[Modul-Kennnummer ]		
Mandatory or elective Module	Elective				<u> </u>			
Creditpoints (LP) and workload	6 LP = 18	0 h						
Module duration	1 Semest	er						
(according to course plan)	2 001100	<b>c.</b>				1		
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Self Study	Creditpoints				
a) Lecture "Plant Physiology"	L	1 - 3 (1 - 3)	138 h	6				
In order to complete the module, yo	u have to	fulfil the following	requirements:					
Compulsory Attendance						20		
Active participation	According	to § 5 para. 3			X	0		
Coursework					5			
Module examination	Usually w	ritten exam (60 mir	ı), alternatively	oral exam (30	min)			
Qualification Goals, learning outcom	e, compet	ences						
<ul> <li>Functions of the comp</li> <li>primary and secondary</li> <li>photosynthetic and dis</li> <li>formation, transport of</li> <li>Uptake and transport of</li> <li>Metabolic cycles (espe</li> <li>Structure and function</li> <li>Regulation of plant def</li> <li>Light receptors, photo</li> </ul>	y reactions ssimilatory storage and of minerals cially nitro of enzyme velopment	of photosynthesis; energy metabolism d mobilisation of as s ggen cycle) es c, hormones, seed g	similates; lipid, ermination; pla	protein and ca nt cancer	rbohydrate me	etabolism;		
Water balance, water	transport a	and plant nutrition						
Compulsory entrance requirements								
Recommended participation require and/or individual courses of the mode and/or individual courses of the mode and a second se		or the module						
Language(s) of instruction and exam			German					
Weight of the module grade in the o		le	Not graded					
Frequency of module offer			Every term					
Reasons for compulsory attendance								
Person responsible for the module			UnivProf. Dr.	Andreas Wach	iter			
Transferability of the module to oth	er degree	programs	Bachelor of Science Biomedical Chemistry, Bachelor of Science Molecular Biotechnology					
Other								



Module EC	Electro	chemistry	[Modul-K	[Modul-Kennnummer ]				
Mandatory or elective Module	Elective				<u>.</u>			
Creditpoints (LP) and workload	6 LP = 18	0 h						
Module duration (according to course plan)	1 Semest	er						
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	when starting in         Mandatory/         Contact           Winter term         elective         Time (SWH)         Self Study         Ci					
a) Lecture "Electrochemistry"	L	2 (1)	М	4	138 h	6		
In order to complete the module, you have to fulfil the following requirements:								
Compulsory Attendance					(	20		
Active participation	According	to § 5 para. 3			× '	0		
Coursework					5	•		
Module examination	Usually w	ritten exam (120 mi	in), alternatively	v oral exam (3	0 min)			
Qualification Goals, learning outcom	e, compet	ences						
<ul> <li>significance.</li> <li>are able to independently work out and deepen contents from the subject area of electrochemistry.</li> <li>have developed an awareness of the connections and links between topics and contents within this highly interdisciplinary field.</li> <li>Contents <ul> <li>Physical basics and terms (conductivity in ionic systems; potentials and structures at phase boundaries; potentials and currents).</li> <li>Electrode materials, electrolyte science, mediators, separators and cell geometries; cyclic voltammetry, spectroelectrochemistry, Marcus theory</li> <li>Corrosion, electrochemical milling and machining; electroplating/metal deposition</li> <li>Production of basic inorganic chemicals</li> <li>Cathode reactions (mediated systems, direct methods, technical applications)</li> <li>Anode reactions (couplings, fluorination, modern concepts)</li> <li>Natural product synthesis</li> <li>Technical electroorganic synthesis</li> <li>Electrochemical surface treatment</li> </ul> </li> </ul>								
<ul> <li>Electropolymerisation,</li> <li>Ion exchangers</li> <li>Bioelectrochemistry, e</li> </ul>								
Compulsory entrance requirements								
Recommended participation require and/or individual courses of the mod		or the module						
Language(s) of instruction and exam	ination		German or Eng	lish				
Weight of the module grade in the o	verall grad	le	Not graded					
Frequency of module offer			Only in the sun	nmer term				
Reasons for compulsory attendance								
Person responsible for the module			N.N.					
Transferability of the module to othe	er degree	programs	Master of Scier	nce Chemistry				
Other				,				

Module APP	Integra	ted Analytical-P	reparative La	ab Course	[Modul-K	(ennnummer ]
Mandatory or elective Module	Elective				<u>_</u>	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration	1 Somost	or				
(according to course plan)	1 Semest			1		T
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	) Self Study Creditpoir	
a) Lecture "Analytical Methods"	L	1 o. 2 (1 o. 2)	М	1	34,5 h	1,5
b) Analytical Preparative Lab Course	APr	1 o. 2 (1 o. 2)	М	9	40,5 h	4,5
In order to complete the module, yo	u have to	fulfil the following	requirements:			
Compulsory Attendance	APr					
Active participation	According	g to § 5 para. 3				
Coursework					0	
Module examination					-	
Qualification Goals, learning outcom	e, compet	tences				
<ul> <li>research using common analytical procedures and, if applicable, isotope labelling.</li> <li>The students are able to: <ul> <li>work independently on research-related topics in preparative organic chemistry and to examine and critically evaluate the results of their work by analytical methods.</li> <li>work out and plan their experiments and implement them independently,</li> <li>debate with their supervisors the performance of the experiments and the analytical techniques used and to correct them,</li> <li>work out solutions when dealing with scientific problems and combine practice and theory,</li> <li>assess the safety aspects of chemicals and experiments and take appropriate action,</li> <li>develop their English language skills through English-language literature and supervisors,</li> <li>work responsibly in a team and to handle hazardous substances,</li> <li>analyse and correct experimental results based on theoretical knowledge through technical literature,</li> <li>assess and optimise the results of experiments and measurements.</li> </ul> </li> <li>Contents</li> <li>Preparation of 2-4 research-related preparations of 1-4 steps in size, 6-8 steps in total. The preparation instructions are taken e.g. from current chemical journals or Organic Syntheses. The obtained pure substances or substance mixtures are analysed with the analytical methods presented in the block lecture, among others, and the results are discussed in the protocol.</li> </ul>						
Depending on the preparation, labelli Compulsory entrance requirements	-					
Recommended participation require and/or individual courses of the modest terms and the modest of th		or the module	Module "Practi	cal Course on	Molecular Synt	hesis"
Language(s) of instruction and exam	ination		German or Eng	lish		
Weight of the module grade in the o	verall grad	de	Not graded			
Frequency of module offer			Every term			
Reasons for compulsory attendance			According to H	ochSchG § 26	Para. 2 (7), Pra	ctical Course
Person responsible for the module			apl. Prof. Dr. H		.,, -	
Transferability of the module to othe	er degree	programs	Master of Scier			
Other			Recommended Organic Synthe	Literature:		en-Weyl

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	Electrons in Molecules					[Modul-Kennnummer ]		
Mandatory or elective Module	Elective				<u> </u>			
Creditpoints (LP) and workload	6 LP = 18	0 h						
Module duration (according to course plan)	1 Semest	er						
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints		
a) Lecture "Electrons in Molecules"	L	1 (2)	М	3	103 <i>,</i> 5 h	4,5		
b) Supporting exercise to a)	E	1 (2)	М	1	34,5 h	1,5		
In order to complete the module, yo	u have to	fulfil the following	requirements:					
Compulsory Attendance					X	0		
Active participation	According	g to § 5 para. 3			5			
Coursework								
Module examination	Usually w	ritten exam (120 m	in), alternatively	y oral exam (3	0 min)			
Qualification Goals, learning outcom	e, compe	tences						
<ul> <li>can establish connection</li> </ul>	ons and lir	s to unknown tasks, iks between topics a		thin the subje	ct and with rel	ated sub-		
disciplines, have gained an unders sciences. Contents	tanding of	the significance of	and contents wi	ucture of mole	ecular systems	in the natural		
disciplines, have gained an unders sciences.	tanding of cules or co bio-inorga Electron to	the significance of ordination compou nic chemistry, spin ransfer in discrete a	the electron str nds with one or crossover comp nd conductivity	more parama pounds, single in extended s	ecular systems gnetic centres molecule mag ystems. Applic	in the natural . Basic nets. Electrical		
disciplines, have gained an unders sciences. Contents Magnetic properties of organic molec concepts, application examples from properties of molecular compounds:	tanding of cules or co bio-inorga Electron to	the significance of ordination compou nic chemistry, spin ransfer in discrete a	the electron str nds with one or crossover comp nd conductivity	more parama pounds, single in extended s	ecular systems gnetic centres molecule mag ystems. Applic	in the natural . Basic nets. Electrical		
disciplines, have gained an unders sciences. Contents Magnetic properties of organic molec concepts, application examples from properties of molecular compounds: from bio-inorganic chemistry. Introdu	tanding of cules or co bio-inorga Electron to nction to n ment(s) fo	the significance of ordination compou unic chemistry, spin ransfer in discrete a nolecular spintronic	the electron str nds with one or crossover comp nd conductivity	more parama pounds, single in extended s	ecular systems gnetic centres molecule mag ystems. Applic	in the natural . Basic nets. Electrical		
disciplines, have gained an unders sciences. Contents Magnetic properties of organic molec concepts, application examples from properties of molecular compounds: from bio-inorganic chemistry. Introdu Compulsory entrance requirements Recommended participation require	tanding of cules or co bio-inorga Electron to nction to n ment(s) fo dule	the significance of ordination compou unic chemistry, spin ransfer in discrete a nolecular spintronic	the electron str nds with one or crossover comp nd conductivity	ucture of mole more parama bounds, single in extended s ifficient data p	ecular systems gnetic centres molecule mag ystems. Applic	in the natural . Basic nets. Electrical		
disciplines, have gained an unders sciences. Contents Magnetic properties of organic molec concepts, application examples from properties of molecular compounds: from bio-inorganic chemistry. Introdu Compulsory entrance requirements Recommended participation require and/or individual courses of the mod	tanding of cules or co bio-inorga Electron tr inction to n ment(s) fo fule ination	iks between topics a the significance of ordination compou nic chemistry, spin ransfer in discrete a nolecular spintronic or the module	the electron str nds with one or crossover comp nd conductivity s for resource-e	ucture of mole more parama bounds, single in extended s ifficient data p	ecular systems gnetic centres molecule mag ystems. Applic	in the natural . Basic nets. Electrical		
disciplines, have gained an unders sciences. Contents Magnetic properties of organic molec concepts, application examples from properties of molecular compounds: from bio-inorganic chemistry. Introdu Compulsory entrance requirements Recommended participation require and/or individual courses of the moo Language(s) of instruction and exam	tanding of cules or co bio-inorga Electron tr inction to n ment(s) fo fule ination	iks between topics a the significance of ordination compou nic chemistry, spin ransfer in discrete a nolecular spintronic or the module	the electron str nds with one or crossover comp nd conductivity s for resource-e German or Eng	ucture of mole more parama bounds, single in extended s fficient data p	ecular systems gnetic centres molecule mag ystems. Applic	in the natural . Basic nets. Electrical		
disciplines, have gained an unders sciences. Contents Magnetic properties of organic molec concepts, application examples from properties of molecular compounds: from bio-inorganic chemistry. Introdu Compulsory entrance requirements Recommended participation require and/or individual courses of the moo Language(s) of instruction and exam Weight of the module grade in the o	tanding of cules or co bio-inorga Electron tr inction to n ment(s) fo fule ination	iks between topics a the significance of ordination compou nic chemistry, spin ransfer in discrete a nolecular spintronic or the module	the electron str nds with one or crossover comp nd conductivity s for resource-e German or Eng Not graded	ucture of mole more parama bounds, single in extended s fficient data p	ecular systems gnetic centres molecule mag ystems. Applic	in the natural . Basic nets. Electrical		
disciplines, have gained an unders sciences. Contents Magnetic properties of organic molec concepts, application examples from properties of molecular compounds: from bio-inorganic chemistry. Introdu Compulsory entrance requirements Recommended participation require and/or individual courses of the moo Language(s) of instruction and exam Weight of the module grade in the o Frequency of module offer	tanding of cules or co bio-inorga Electron tr inction to n ment(s) fo fule ination	iks between topics a the significance of ordination compou nic chemistry, spin ransfer in discrete a nolecular spintronic or the module	the electron str nds with one or crossover comp nd conductivity s for resource-e German or Eng Not graded	ucture of mole more parama pounds, single in extended s fficient data p lish	ecular systems gnetic centres molecule mag ystems. Applic rocessing.	in the natural . Basic nets. Electrical		
disciplines, have gained an unders sciences. Contents Magnetic properties of organic molec concepts, application examples from properties of molecular compounds: from bio-inorganic chemistry. Introdu Compulsory entrance requirements Recommended participation require and/or individual courses of the moo Language(s) of instruction and exam Weight of the module grade in the o Frequency of module offer Reasons for compulsory attendance	tanding of cules or co bio-inorga Electron to nent(s) fo dule ination verall grad	iks between topics a the significance of ordination compou unic chemistry, spin ransfer in discrete a nolecular spintronic or the module	and contents wi the electron str nds with one or crossover comp nd conductivity s for resource-e German or Eng Not graded Only in the win	ucture of mole more parama bounds, single in extended s fficient data p lish ter term	ecular systems gnetic centres molecule mag ystems. Applic rocessing.	in the natural . Basic nets. Electrical		

Module SK	Supram	olecular Cataly	alysis [Modul-Kennnumm				
Mandatory or elective Module	Elective					<u></u>	
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self	f Study	Creditpoints
a) Lecture "Supramolecular Catalysis"	L	2 (1)	М	3	103 <i>,</i> 5 h		4,5
b) Supporting exercise to a)	E	2 (1)	М	1	34	4,5 h	1,5
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance							
Active participation	According	g to § 5 para. 3					
Coursework					. 0		
Module examination	Usually w	ritten exam (120 m	in), alternativel	y oral exam (3	0 min)		
Qualification Goals, learning outcom	ie, compe	tences					
<ul> <li>can describe the basic</li> <li>can work out and deep</li> <li>can establish connectidisciplines,</li> <li>can transfer the learner</li> </ul>	pen partial ons and lir	contents independ iks between topics	and contents wi	thin the subje	ct and	with rela	ated sub-
Contents							
Use of supramolecular interactions ar and light-driven catalyses are discuss interactions and resulting catalytic ac confinement, inhibition and feedback effects in colloids, polymers and solic	ed with er tivity. Cat loops, an	nphasis on the corre alysis mechanisms i d autocatalysis are	elation betweer ncluding enanti discussed. For h	n catalyst struc oselective cata neterogeneous	cture, s alysis, (	supramol catalysis	lecular under spatial
Compulsory entrance requirements							
Recommended participation require and/or individual courses of the modest courses courses of the modest courses courses of the modest courses of the modes		or the module					
Language(s) of instruction and exam	ination		German or Eng	lish			
Weight of the module grade in the o	verall grad	de	Not graded				
Frequency of module offer			Only in the sun	nmer term			
Reasons for compulsory attendance							
Person responsible for the module			UnivProf. Dr.	Carsten Streb			
Transferability of the module to oth	er degree	programs	Master of Scier	nce Chemistry			
Other							

Module MPC	Molecu	Molecular Photochemistry				
Mandatory or elective Module	Elective				<u> </u>	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture "Molecular Photochemistry"	L	1 (2)	М	3	103,5 h	4,5
b) Supporting exercise to a)	E	1 (2)	М	1	34,5 h	1,5
In order to complete the module, y	ou have to	fulfil the following	requirements:			
Compulsory Attendance						
Active participation	According	to § 5 para. 3				
Coursework					0	
Module examination	Usually w	ritten exam (120 m	in), alternatively	/ oral exam (3	0 min)	
Qualification Goals, learning outco	me, compet	ences				
disciplines, are able to transfer th gain a comprehensive fundamental concept	e overview o	of the cross-section			y, with an equ	
Electron transfer, fundamentals of p chromophores, photokinetics, optic	al spectrosc	stry, photophysics a opy, photocatalysis	and photochemi	onversion, nat	ural and artific	organic
Electron transfer, fundamentals of p chromophores, photokinetics, optic	al spectrosc bes, supram	stry, photophysics a opy, photocatalysis olecular photochen	and photochemi s, solar energy c nistry, organic p	onversion, nat	ural and artific	organic
photosynthesis, photochemical prol	al spectrosc bes, supram hotochemist	stry, photophysics a opy, photocatalysis olecular photochen	and photochemi s, solar energy c nistry, organic p	onversion, nat	ural and artific	organic
Electron transfer, fundamentals of p chromophores, photokinetics, optic photosynthesis, photochemical prol rearrangements, fragmentations, pl Compulsory entrance requirement: Recommended participation require	al spectrosc bes, supram hotochemist s rement(s) fo	stry, photophysics a opy, photocatalysis olecular photochen try in biological syst	and photochemi s, solar energy c nistry, organic p	onversion, nat	ural and artific	organic
Electron transfer, fundamentals of p chromophores, photokinetics, optic photosynthesis, photochemical prol rearrangements, fragmentations, pl Compulsory entrance requirement Recommended participation requir and/or individual courses of the mo	al spectrosc bes, supram hotochemist s rement(s) fo odule	stry, photophysics a opy, photocatalysis olecular photochen try in biological syst	and photochemi s, solar energy c nistry, organic p	onversion, nat	ural and artific	organic ial
Electron transfer, fundamentals of p chromophores, photokinetics, optic photosynthesis, photochemical pro- rearrangements, fragmentations, pl Compulsory entrance requirements Recommended participation require and/or individual courses of the me Language(s) of instruction and examples	al spectrosc bes, supram hotochemist s rement(s) fo odule mination	stry, photophysics a opy, photocatalysis olecular photochen try in biological syst	and photochemi s, solar energy c nistry, organic p ems.	onversion, nat	ural and artific	organic ial
Electron transfer, fundamentals of p chromophores, photokinetics, optic photosynthesis, photochemical prol rearrangements, fragmentations, pl Compulsory entrance requirements Recommended participation requir and/or individual courses of the m Language(s) of instruction and exam Weight of the module grade in the	al spectrosc bes, supram hotochemist s rement(s) fo odule mination	stry, photophysics a opy, photocatalysis olecular photochen try in biological syst	and photochemi s, solar energy c nistry, organic p ems. German or Eng	onversion, nat hotoreactions	ural and artific	organic ial
Electron transfer, fundamentals of p chromophores, photokinetics, optic photosynthesis, photochemical prof rearrangements, fragmentations, pl Compulsory entrance requirements Recommended participation require and/or individual courses of the mo Language(s) of instruction and exam Weight of the module grade in the Frequency of module offer	al spectrosc bes, supram hotochemist s rement(s) fo odule mination overall grad	stry, photophysics a opy, photocatalysis olecular photochen try in biological syst	and photochemi s, solar energy c nistry, organic p ems. German or Eng Not graded	onversion, nat hotoreactions	ural and artific	organic ial
Electron transfer, fundamentals of p chromophores, photokinetics, optic photosynthesis, photochemical prol rearrangements, fragmentations, pl Compulsory entrance requirement: Recommended participation requir and/or individual courses of the me Language(s) of instruction and exar Weight of the module grade in the Frequency of module offer Reasons for compulsory attendance	al spectrosc bes, supram hotochemist s rement(s) fo odule mination overall grad	stry, photophysics a opy, photocatalysis olecular photochen try in biological syst	and photochemi s, solar energy c nistry, organic p ems. German or Eng Not graded	onversion, nat hotoreactions lish ter term	ural and artific	organic ial
Electron transfer, fundamentals of p chromophores, photokinetics, optic photosynthesis, photochemical prol rearrangements, fragmentations, pl	al spectrosc bes, supram hotochemist s rement(s) fo odule mination overall grad	stry, photophysics a opy, photocatalysis olecular photochen try in biological syst or the module	and photochemi s, solar energy co nistry, organic p ems. German or Eng Not graded Only in the win	onversion, nat hotoreactions lish ter term Katja Heinze	ural and artific	organic ial

Module Handbook	Version	As of	JG
Master of Science Biomedical Chemistry	2.0	14.03.2024	JGIO

Module FMM	Advanced Laboratory Course on Functional [Modul-Kennnum Molecular Materials					(ennnummer ]	
	Molecu	llar Materials					
Mandatory or elective Module	Elective						
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er				<b>.</b>	
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lab Course "Functional Molecular Materials"	APr	2 (1)	М	9	40,5 h	4,5	
b) Supporting Seminar to a)	S	2 (1)	М	1	34,5 h	1,5	
In order to complete the module, yo	u have to	fulfil the following	requirements:				
Compulsory Attendance	APr, S						
Active participation	According	g to § 5 para. 3			0		
Coursework							
Module examination				<b>\O</b>			
Qualification Goals, learning outcom	e, compe	tences					
evaluate them and ass are proficient in the th according to the rules are able to handle haz environmental regulat Contents	eoretical I of good sc ardous sul	background of their ientific practice	$\sim$				
Conducting experiments to elaborate and analytical methods, e.g. investiga experiments, determination of turno experimental results with theoretical	ntion of ele ver curves	ectronic and magne of catalyses or pho	tic properties, lu	uminescence c	or time-resolve	d spectroscopic	
Compulsory entrance requirements							
Recommended participation require and/or individual courses of the modest the modest set to be a set to		or the module	Modules "Molecular Photochemistry", "Supramolecular Catalysis" und "Electrons in Molecules"				
Language(s) of instruction and exam	ination		German or English				
Weight of the module grade in the o	verall grad	de	Not graded				
Frequency of module offer			Only in the summer term				
Reasons for compulsory attendance			According to HochSchG § 26 Abs. 2 (7), internship; internship-accompanying upper seminar according to § 5 Abs. 5: discussion of safety-relevant details of and discussion of internship experiments.				
Person responsible for the module			UnivProf. Dr. Carsten Streb				
Transferability of the module to othe	er degree	programs	Master of Scier	nce Chemistry			
Other							



Module SpA	Trace A	nalysis I			[Modul-K	ennnummer ]
Mandatory or elective Module	Elective				<u>_</u>	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture "Inorganic Trace and Species Analysis"	L	1 (2)	М	2	69 h	3
b) Lecture "Organic Trace Analysis"	L	1 (2)	М	2	69 h	3
In order to complete the module, yo	ou have to	fulfil the following	requirements:			
Compulsory Attendance						
Active participation	Gemäß §	5 Abs. 3				
Coursework	1				3	
Module examination	Usually w and b)	ritten exam (120 m	in), alternatively	v oral exam (3	0 min) on the co	ontents of a)
Qualification Goals, learning outcom	ne, compe	tences				
<ul> <li>knowledge.</li> <li>The students are able to: <ul> <li>reproduce principles f</li> <li>identify the main area analysis, species analy</li> <li>relate keywords such a detection to the meth</li> <li>evaluate analytical meto a set trace analytica</li> </ul> </li> </ul>	s of applica sis, medica as food saf ods used ethods and	ation of analysis, su al and diagnostic an ety or water contan	ch as environme alysis nination, doping	ental analysis, g tests, geneti	technical and ir c analysis or aut	henticity
<ul> <li>to a set trace analytic</li> <li>to understand the exp critically evaluate this</li> </ul>		lished in analytical t	extbooks as we	ll as in interna	tional journals	and to
to understand the exp critically evaluate this Contents	material					
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Module SpaP	Trace A	race Analysis II					ennnummer ]
Mandatory or elective Module	Elective				<u> </u>		
Creditpoints (LP) and workload	6 LP = 180	5 LP = 180 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self S	tudy	Creditpoints
a) Practical course "Trace Analysis II"	APr	2 (1)	М	4	78	h	4
b) Supporting Seminar to a)	Seminar	2 (1)	М	2	39	h	2
In order to complete the module, yo	u have to	fulfill the following	requirements:	•			·
Compulsory Attendance	APr, S					X	0
Active participation	According to § 5 para. 3						
Coursework							
Module examination					0		
Qualification Goals, learning outcom	e, compet	ences					

Building on analytical contents and working techniques already learned in the Bachelor's degree programme, students acquire special expertise in the field of advanced instrumental trace analysis in the module Instrumental Trace Analysis II (practical course). The contents are developed, deepened and practically implemented in the form of an advanced practical course on organic trace analysis and elemental analysis and a lecture seminar. Newly acquired knowledge is always integrated into the existing knowledge. The students acquire in-depth knowledge of the current methods of instrumental trace analysis (chromatography, atomic spectrometry, molecular spectroscopy, mass spectrometry).

The students are able to:

- apply advanced analytical-instrumental working techniques
- statistically evaluate recorded measurement data
- carry out trace analysis work independently and on their own responsibility
- scientifically record, interpret and present the results of their experiments
- agree on individual work steps when working in groups of two, to plan them together and to implement them in a coordinated manner
- realise demanding research-related experiments in parallel within a time window (self-, time- and resource management)
- analyse and evaluate current scientific literature
  - independently prepare and present a scientific Presentation on a (given) current analytical-chemical topic.

#### Contents

a) Experiments in groups of two on the determination of organic analytes by means of GC-MS and HPLC-MS (mode of operation, set-up, column types, ionisation techniques, detectors, analysers, MS/MS, derivatisation), by means of ambient MS (set-up and mode of operation of corresponding ion sources, advantages and disadvantages, areas of application), and by means of aerosol mass spectrometry (AMS). Experiments in groups of two on inorganic trace analysis based on analyte samples of different matrices by means of mass and emission spectrometry in connection with inductively coupled plasma (ICP-OES, ICP-MS) and X-ray spectroscopy (TXRF). Consideration of different sample preparation/digestion methods and systems of sample introduction.

b) Current analytical-chemical topics are discussed. The students independently prepare a scientific presentation on one of these given topics and present it within the framework of the seminar. Independent research and evaluation of relevant literature are important.

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Compulsory entrance requirements	Module "Trace Analysis I"
Recommended participation requirement(s) for the module and/or individual courses of the module	
Language(s) of instruction and examination	German or English
Weight of the module grade in the overall grade	Not graded
Frequency of module offer	Only in the summer term

Module Handbook	Version	As of	JG
Master of Science Biomedical Chemistry	2.0	14.03.2024	JGIO

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Description       Description       Description       Description         a) Lecture       Part 1: "Synthesis and use of polymers".       L       1 - 3 (1 - 3)       M       3       103,5 h       4,5         Part 2: "Physical Chemistry of Polymers".       L       1 - 3 (1 - 3)       M       1       34,5 h       1,5         In order to complete the module, you have to fulfil the following requirements:       Compulsory Attendance       Image: Compulsory Attendance       Image: Compulsory Attendance         Active participation       b) According to § 5 para. 3 (usually exercise assignments)       Coursework       Image: Compulsory Attendance         Module examination       Usually written exam (120 min), alternatively oral exam (30 min)       Image: Compulsory Attendance         Qualification Goals, learning outcome, competences       Image: Compulsory Attendance       Image: Compulsory Attendance         Active participation       Usually written exam (120 min), alternatively oral exam (30 min)       Image: Computer Material as used as the central methods of polymer characterisation and step growth. An overview of relevant polymer materials as well as the central methods of polymer characterisation and basic properties of polymers in comparison to other material classes, especially to low-molecular compounds.       Image: Comparison to other material classes, especially to low-molecular compounds.       Image: Comparison to other material classes, especially to low-molecular compounds.       Image: Comparison to ther material c	Module MC1	Macror	nolecular Chem	istry		[Modu	ll-Kennnummer ]
Module duration (according to course plan)         1 Semester           Courses/ Learning formats         Type         Regular term (Summer term)         Mandatory/ elective         Contact Time (SWH)         Self Study         Creditpo           a) Lecture Part 1: "Synthesis and use of polymers".         L         1 - 3 (1 - 3)         M         3         103,5 h         4,5           Part 2: "Physical Chemistry of Polymers".         L         1 - 3 (1 - 3)         M         1         34,5 h         1,5           In order to complete the module, you have to fulfil the following requirements:         Compulsory Attendance         Coursework         Coursework is in the solid state is taught.         The students acquire the basics of polymer chemistry, types of polymer characterisation and basic properties of polymers in comparison to other material classes, especially to low-molecular compounds.         e acquire the basics of polymer chemistry, types of polymerisation, chain and step growth,         ertiteally evaluate polymerisation methods, bot with regard to the achievable molecular w	Mandatory or elective Module	Elective				<u> </u>	
Isemester         Courses/ Learning formats       Type       Regular term when starting in (Summer term)       Mandatory/ elective       Contact Time (SWH)       Self Study       Creditpo         a) Lecture Part 1: "Synthesis and use of polymers".       L       1 - 3 (1 - 3)       M       3       103,5 h       4,5         Part 2: "Physical Chemistry of Polymers".       E       1 - 3 (1 - 3)       M       1       34,5 h       1,5         In order to complete the module, you have to fulfil the following requirements:       Compulsory Attendance       Active participation       b) According to § 5 para. 3 (usually exercise assignments)       Coursework         Module examination       Usually written exam (120 min), alternatively oral exam (30 min)       Qualification Goals, learning outcome, competences       The students acquire the basics of polymer chemistry, types of polymer characterisation and basic properties of polymers ir solution as well as in the solid state is taught.         The students are able to: <ul> <li>reproduce basic physical properties and material properties of polymers and special features of polymers ir somparison to other material classes, especially to low-molecular compounds.</li> <li>acquire the basics of polymer chemistry, types of polymerisation, chain and step growth,</li> <li>critically evaluate polymerisation methods, both with regard to the achievable molecular weights and with regard to the respective limitations concerning polydispersity,</li> <li>get to know basic characterisation methods, both with reg</li></ul>	Creditpoints (LP) and workload	6 LP = 18	0 h				
Courses/ Learning formatsTypewhen starting in Winter term (Summer term)Mandatory/ electiveContact Time (SWH)Self StudyCreditpoa) Lecture Part 1: "Synthesis and use of polymers".L1 - 3 (1 - 3)M3103,5 h4,5Part 2: "Physical Chemistry of Polymers".L1 - 3 (1 - 3)M134,5 h1,5In order to complete the module, you have to fulfil the following requirements:Compulsory AttendanceCourseworkCourseworkActive participationb) According to § 5 para. 3 (usually exercise assignments)CourseworkCourseworkModule examinationUsually written exam (120 min), alternatively oral exam (30 min)Qualification Goals, learning outcome, competencesThe students acquire the basics of polymer chemistry, types of polymer characterisation and basic properties of polymers in solution as well as in the solid state is taught.In order to compounds.• reproduce basic physical properties and material properties of polymers and special features of polymers in comparison to other material classes, especially to low-molecular compounds.• reproduce basic characterisation methods, both with regard to the achievable molecular weights and with regard to the respective limitations concerning polydispersity, • get to know basic characterisation methods and to evaluate them with regard to their suitability for specific questions• conceptually understand and quantitatively discuss the structure and dynamics of macromolecules and to evaluate the suitability for specific questions		1 Semest	ter				
Part 1: "Synthesis and use of polymers". L 1 - 3 (1 - 3) M 3 103,5 h 4,5 Part 2: "Physical Chemistry of Polymers". L 1 - 3 (1 - 3) M 1 34,5 h 1,5 In order to complete the module, you have to fulfil the following requirements: Compulsory Attendance Active participation b) According to § 5 para. 3 (usually exercise assignments) Coursework M 400000000000000000000000000000000000		Туре	when starting in Winter term			Self Study	Creditpoints
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Compulsory Attendance           Active participation         b) According to § 5 para. 3 (usually exercise assignments)           Coursework         Module examination <b>Qualification Goals, learning outcome, competences</b> The students acquire the basics of polymer chemistry, types of polymerisation, chain and step growth. An overview of relevant polymer materials as well as the central methods of polymer characterisation and basic properties of polymers ir solution as well as in the solid state is taught.           The students are able to:         • reproduce basic physical properties and material properties of polymers and special features of polymers ir comparison to other material classes, especially to low-molecular compounds.           • acquire the basics of polymer chemistry, types of polymerisation, chain and step growth,           • critically evaluate polymerisation methods, both with regard to the achievable molecular weights and with regard to the respective limitations concerning polydispersity,           • get to know basic characterisation methods and to evaluate them with regard to their suitability for specific questions	<ul><li>b) Supporting exercise to a)</li></ul>	E	1 - 3 (1 - 3)	М	1	34,5 h	1,5
Active participation       b) According to § 5 para. 3 (usually exercise assignments)         Coursework       Module examination         Usually written exam (120 min), alternatively oral exam (30 min)         Qualification Goals, learning outcome, competences         The students acquire the basics of polymer chemistry, types of polymerisation, chain and step growth. An overview of relevant polymer materials as well as the central methods of polymer characterisation and basic properties of polymers in solution as well as in the solid state is taught.         The students are able to:       • reproduce basic physical properties and material properties of polymers and special features of polymers in comparison to other material classes, especially to low-molecular compounds.         • acquire the basics of polymer chemistry, types of polymerisation, chain and step growth,         • critically evaluate polymerisation methods, both with regard to the achievable molecular weights and with regard to the respective limitations concerning polydispersity,         • get to know basic characterisation methods and to evaluate them with regard to their suitability for specific questions         • conceptually understand and quantitatively discuss the structure and dynamics of macromolecules and to	In order to complete the module, y	ou have to	fulfil the following	requirements:			
Coursework         Usually written exam (120 min), alternatively oral exam (30 min)           Qualification Goals, learning outcome, competences         State           The students acquire the basics of polymer chemistry, types of polymerisation, chain and step growth. An overview of relevant polymer materials as well as the central methods of polymer characterisation and basic properties of polymers ir solution as well as in the solid state is taught.           The students are able to:         • reproduce basic physical properties and material properties of polymers and special features of polymers ir comparison to other material classes, especially to low-molecular compounds.           • acquire the basics of polymer chemistry, types of polymerisation, chain and step growth,         • critically evaluate polymerisation methods, both with regard to the achievable molecular weights and with regard to the respective limitations concerning polydispersity,           • get to know basic characterisation methods and to evaluate them with regard to their suitability for specific questions	Compulsory Attendance					0	
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	Contents						

#### Part 1:

General basics: tasks of polymer science, polymer structures, nomenclature.

Polymer synthesis: Polycondensation (step growth), Carothers equation, polymerisations with chain growth,

Radical and ionic methods of polymer synthesis, kinetics, chain transfer, copolymerisation, catalytic polymerisation,

polyinsertion

Polymerisation, polyinsertion, catalysts (initiators).

Polymerisation in heterophase (emulsion, dispersion, suspension).

Polymer modification: cellulose, rubber, polymer analogue reactions.

Controlled and living polymerisation processes, ring opening reactions, solid phase synthesis.

Part 2:

Polymer structure: block copolymers, conformation of macromolecules, errant statistics, RIS model, ideal and real chain statistics, entropy elasticity, Flory exponent and scale laws.

Molecular characterisation of polymers in solution: colligative methods, gel permeation chromatography,

mass spectrometry, static light scattering.

Polymer dynamics: Rouse and Zimm model.

Polymer thermodynamics: Flory-Huggins theory, phase diagrams.

Compulsory entrance requirements	
Recommended participation requirement(s) for the module and/or individual courses of the module	
Language(s) of instruction and examination	English
Weight of the module grade in the overall grade	Not graded



Frequency of module offer	Every term
Reasons for compulsory attendance	
Person responsible for the module	Prof. Dr. Andreas Walther
Transferability of the module to other degree programs	Bachelor of Science Biomedical Chemistry, Bachelor of Science Chemistry, Master of Science Chemistry
Other	Recommended Literature: Tieke – Makromolekulare Chemie. Eine Einführung (Wiley). Koltzenburg, Maskos, Nuyken – Polymere: Synthese, Eigenschaften und Anwendungen (Springer) Lechner, Gehrke, Nordmeier – Makromolekulare Chemie (Springer) Seiffert – Physical Chemistry of Polymers: A Conceptual Introduction (DeGruyter)
hormation	noute



Module MC1P	Practic	al Course Bioma	cromolecula	r Chemistry	[Modul-I	Kennnummer ]	
Mandatory or elective Module	Elective				<u> </u>		
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	ter					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Practical Course Biomacromolecular Chemistry for Advanced Students 1	APr	1 - 3 (1 - 3)	м	6	117 h	6	
In order to complete the module, ye	ou have to	fulfil the following	requirements:				
Compulsory Attendance	APr					,	
Active participation	Accordin	g to § 5 para. 3					
Coursework					2		
Module examination					0		
Qualification Goals, learning outcor	ne, compe	tences					
<ul> <li>produce biomedically</li> <li>deal effectively with t within a defined time</li> </ul>	heir time a		nning work pro	cesses indeper	ndently and rea	alising them	
Practical experiments are selected fr Experiments on polymer synthesis: r copolymerisation, polymerisation in biopolymers, silicones and biodegrad	adical poly heteropha	merisation, polycon se, networks; as we					
Compulsory entrance requirements	i i		Module "Macr	omolecular Ch	emistry		
Recommended participation requir and/or individual courses of the mo		or the module					
Language(s) of instruction and exan	nination		German or English				
Weight of the module grade in the	overall gra	de	Not graded				
Frequency of module offer			Every term				
Reasons for compulsory attendance			According to HochSchG § 26 Para. 2 (7), Practical Course				
Person responsible for the module			UnivProf. Dr.	Sebastian Seif	fert		
Transferability of the module to oth	ner degree	programs	Bachelor of Sci	ence Biomedic	al Chemistry		
Other							

Module Handbook	Version	As of	JG
Master of Science Biomedical Chemistry	2.0	14.03.2024	JGIO

Module MC2	D.C. alla m				[Modul-K	ennnummer ]
	Materia	n and Industrial als	Aspects of P	olymer		enninumner j
Mandatory or elective Module	Elective				<u> </u>	
Creditpoints (LP) and workload	6 LP = 180 h					
Module duration						
(according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture Part 1: "Synthesis and Use of Polymer Materials" Part 2: "Physical Chemistry of Polymeric Materials	L	1 (2)	М	3	103,5 h	4,5
b) Seminar "Modern and Industrial Aspects of Polymer Materials"	S	1 (2)	М	1	34,5 h	1,5
In order to complete the module, yo	ou have to	fulfil the following	requirements:			
Compulsory Attendance				.0		
Active participation	According	g to § 5 para. 3 (usua	ally successful p	resentation ir	the seminar)	
Coursework						
Module examination	Usually w	ritten exam (120 m	in), alternatively	/ oral exam (3	0 min)	
Qualification Goals, learning outcom	ne, compe	tences				
understand current re elastomers, composite and bio-inspired mate edescribe the rheology	e materials rial design	, weak interactions				
<ul> <li>phenomenologically, b</li> <li>reproduce the basic ch crystals.</li> </ul>	both qualit	rs in the melt and so atively and quantita	itively.	ethodological	ly, conceptually	and
<ul> <li>reproduce the basic ch crystals.</li> </ul>	both qualit	rs in the melt and so atively and quantita	itively.	ethodological	ly, conceptually	and
<ul> <li>reproduce the basic ch crystals.</li> <li>Contents</li> <li>Modern methods of polymer synthes         <ul> <li>Advanced composite r</li> <li>Responsive and switch</li> <li>Biomimetic concepts i</li> <li>Phase-segregated poly</li> <li>Polymer nanoparticles</li> </ul> </li> <li>Fundamentals of rheology:         <ul> <li>viscoelasticity</li> <li>Complex rheological m</li> <li>Time-temperature sup</li> <li>Rheology of polymer s transition.</li> </ul> </li> </ul>	both qualit maracterist sis: materials, l mable mate n polymer ymer syste s and self-a material pro- perposition systems: Ro in each ca	rs in the melt and so atively and quantita ics of the structure a high performance m rials science ms in application, th issembled nanostru operties eptation in melt and se separate treatme	and dynamics o naterials nermoplastic ela ctures	ethodological f polymeric so astomers er elasticity of	ly, conceptually lutions, gels, gla	and asses and
<ul> <li>reproduce the basic ch crystals.</li> <li>Contents</li> <li>Modern methods of polymer synthes         <ul> <li>Advanced composite r</li> <li>Responsive and switch</li> <li>Biomimetic concepts i</li> <li>Phase-segregated poly</li> <li>Polymer nanoparticles</li> </ul> </li> <li>Fundamentals of rheology:         <ul> <li>viscoelasticity</li> <li>Complex rheological n</li> <li>Time-temperature sup</li> <li>Rheology of polymer s transition.</li> </ul> </li> <li>Building on this: comprehensive and the state of melts, semi-dilute solution</li> </ul>	both qualit maracterist sis: materials, l mable mate n polymer ymer syste s and self-a material pro- perposition systems: Ro in each ca	rs in the melt and so atively and quantita ics of the structure a high performance m rials science ms in application, th issembled nanostru operties eptation in melt and se separate treatme	and dynamics o naterials nermoplastic ela ctures	ethodological f polymeric so astomers er elasticity of	ly, conceptually lutions, gels, gla	and asses and
<ul> <li>reproduce the basic ch crystals.</li> <li>Contents</li> <li>Modern methods of polymer synthes         <ul> <li>Advanced compositer</li> <li>Responsive and switch</li> <li>Biomimetic concepts i</li> <li>Phase-segregated poly</li> <li>Polymer nanoparticles</li> </ul> </li> <li>Fundamentals of rheology:         <ul> <li>viscoelasticity</li> <li>Complex rheological n</li> <li>Time-temperature sup</li> <li>Rheology of polymer s transition.</li> </ul> </li> <li>Building on this: comprehensive and the state of melts, semi-dilute solution</li> <li>Compulsory entrance requirements</li> <li>Recommended participation require</li> </ul>	sis: materials, l mable materials, l mable materials, l mable materials, l mable materials, l materials, l materials, l material pro- perposition systems: Re in each ca ons, gels, c ement(s) for	rs in the melt and so atively and quantita ics of the structure a high performance m rials science ms in application, th issembled nanostru operties eptation in melt and se separate treatme rystals and partial c	and dynamics o naterials nermoplastic ela ctures	ethodological f polymeric so astomers er elasticity of ure, dynamics	ly, conceptually lutions, gels, gla networks, dyna and properties	and asses and
<ul> <li>reproduce the basic ch crystals.</li> <li>Contents</li> <li>Modern methods of polymer synthes         <ul> <li>Advanced compositer</li> <li>Responsive and switch</li> <li>Biomimetic concepts i</li> <li>Phase-segregated poly</li> <li>Polymer nanoparticles</li> </ul> </li> <li>Fundamentals of rheology:         <ul> <li>viscoelasticity</li> <li>Complex rheological m</li> <li>Time-temperature sup</li> <li>Rheology of polymer s transition.</li> </ul> </li> <li>Building on this: comprehensive and the state of melts, semi-dilute solution</li> <li>Compulsory entrance requirements</li> <li>Recommended participation required and/or individual courses of the more</li> </ul>	both qualit maracterist sis: materials, I mable mate n polymer ymer syste s and self-a material pro- perposition systems: Re in each ca ons, gels, c ement(s) for dule	rs in the melt and so atively and quantita ics of the structure a high performance m rials science ms in application, th issembled nanostru operties eptation in melt and se separate treatme rystals and partial c	and dynamics o naterials hermoplastic ela ctures	ethodological f polymeric so astomers er elasticity of ure, dynamics	ly, conceptually lutions, gels, gla networks, dyna and properties	and asses and
<ul> <li>reproduce the basic ch crystals.</li> <li>Contents</li> <li>Modern methods of polymer synthes         <ul> <li>Advanced composite r</li> <li>Responsive and switch</li> <li>Biomimetic concepts i</li> <li>Phase-segregated poly</li> <li>Polymer nanoparticles</li> </ul> </li> <li>Fundamentals of rheology:         <ul> <li>viscoelasticity</li> <li>Complex rheological n</li> <li>Time-temperature sup</li> <li>Rheology of polymer s transition.</li> </ul> </li> <li>Building on this: comprehensive and the state of melts, semi-dilute solution</li> <li>Compulsory entrance requirements</li> <li>Recommended participation require</li> </ul>	both qualit maracterist sis: materials, I mable materials material pro- perposition systems: Re- in each ca- ons, gels, c ement(s) for dule	rs in the melt and so atively and quantita ics of the structure i nigh performance m rials science ms in application, th issembled nanostru operties eptation in melt and se separate treatme rystals and partial c	and dynamics o naterials nermoplastic ela ctures	ethodological f polymeric so astomers er elasticity of ure, dynamics	ly, conceptually lutions, gels, gla networks, dyna and properties	and asses and



f. Dr. Andreas Walther Science Chemistry Inded Literature: Irg, Maskos, Nuyken – Polymere: Synthese, ften und Anwendungen (Springer) Gehrke, Nordmeier – Makromolekulare Chemie h, Colby – Polymer Physics (Oxford University
ended Literature: Irg, Maskos, Nuyken – Polymere: Synthese, ften und Anwendungen (Springer) Gehrke, Nordmeier – Makromolekulare Chemie n, Colby – Polymer Physics (Oxford University
rrg, Maskos, Nuyken – Polymere: Synthese, ften und Anwendungen (Springer) Gehrke, Nordmeier – Makromolekulare Chemie n, Colby – Polymer Physics (Oxford University
evarantee



Module MC3	Colloid Chemistry and Medical Polymers [Modul-Kennn					Kennnummer ]	
Mandatory or elective Module	Elective	Elective					
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lecture "Colloid Chemistry"	L	1 (2)	М	2	69 h	3	
b) Lecture "Medically relevant polymers"	L	1 (2)	М	2	69 h	3	
In order to complete the module,	you have to	fulfil the following	requirements:				
Compulsory Attendance							
Active participation	According	g to § 5 para. 3					
Coursework					0		
Module examination	b) Usually	v written exam (60 n v written exam (60 r ns must be passed,	nin), alternative	ely oral exam (	30 min).	ic mean of both	
Qualification Goals, learning outco	ome, compe	tences					
An in-depth insight into the produc medical applications is provided. The students are able to: • reproduce and expla • discuss colloidal syst • work out and reprod	in methods ems with re	for the investigation gard to their charac	of nanostructu teristic time, ler	ures and (poly ngth and energ	mer) surfaces, gy scales,		
Contents							
<ul> <li>a) Interfacial and colloid chemistry with different properties for different b) Synthesis methods for materials biodegradation of polymeric mater (aliphatic polyesters, polyethylene and vaccines; artificial extracellular</li> </ul>	ent application for use in m ials; biocom glycol, silico	ons, characterisation edicine, implants fo patibility and biode nes, polypeptides an	n. Ir dental applica gradability of po	itions or as pro plymer classes	ostheses; basic for medical ap	principles of oplications	
Compulsory entrance requirement	ts						
Recommended participation requi and/or individual courses of the m		or the module	Module "Macr	omolecular Ch	emistry"		
Language(s) of instruction and exa	mination		German or Eng	lish			
Weight of the module grade in the	overall grad	de	Not graded				
Frequency of module offer			Only in the win	iter term			
Reasons for compulsory attendance	ce						
Person responsible for the module	•		UnivProf. Dr.	Holger Frey			
Transferability of the module to of	har dagraa						
mansferability of the module to of	ther degree	programs	Master of Scier	nce Chemistry			

Module Handbook	Version	As of	JG
Master of Science Biomedical Chemistry	2.0	14.03.2024	JGIO

Module MC4	Comple	x (Supra)Molec	ular Systems	and	[Modul-K	(ennnummer ]
	Biopoly	Biopolymers				
Mandatory or elective Module	Elective				<u>L</u>	
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	er				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Lecture "Complex (Supra)Molecular Systems"	L	2 (1)	м	2	69 h	3
b) Lecture "Biopolymers"	L	2 (1)	М	2	69 h	3
In order to complete the module, yo	u have to	fulfil the following	requirements:			
Compulsory Attendance						
Active participation	According	g to § 5 para. 3			0	
Coursework						
Module examination	<ul> <li>a) Usually written exam (60 min), alternatively oral exam (30 min)</li> <li>b) Usually written exam (60 min), alternatively oral exam (30 min).</li> <li>Both exams must be passed, the module grade results from the arithmetic mean of bot exams.</li> </ul>					
Qualification Goals, learning outcom	ne, compet	tences				
<ul> <li>The students are able to:</li> <li>evaluate biologically reassembly,</li> <li>understand and apply synthetic systems,</li> <li>understand and reprod</li> <li>Distinguish equilibrium</li> <li>Understand the basics</li> </ul>	recognitio duce biolo n and non-	n motifs, weak inter gical and chemical r equilibrium systems	<ul> <li>ractions and org</li> <li>eaction networ</li> <li>s.</li> </ul>	ganisational pr ks and their dy	inciples in natu vnamics.	ral and
Contents						
<ul> <li>a) Supramolecular Chemistry and Sup Networks and Systems; Non-equilibri Adaptive and Interactive Materials.</li> <li>b) Polysaccharides (cellulose and der polyisoprenoids and natural rubber);</li> </ul>	ium States ivatives, ch Nanocellu	; Chemical Reaction nitin, starch, glycoge lose/nanochitin/ba	Networks, Dyn en); Lignins; Poly cterial cellulose	amic DNA Nan vesters (polyhy ; Polynucleotic	oscience, Dissi vdroxyalkanoat des in materials	pative, es), s context (DNA,
RNA); Proteins and scleroproteins (co						
RNA); Proteins and scleroproteins (cc Compulsory entrance requirements						
Compulsory entrance requirements Recommended participation require		or the module	Module "Macro	omolecular Ch	emistry"	
Compulsory entrance requirements Recommended participation require and/or individual courses of the mod	dule	or the module	Module "Macro German or Eng		emistry"	
Compulsory entrance requirements Recommended participation require and/or individual courses of the mod	dule ination				emistry"	
Compulsory entrance requirements Recommended participation require and/or individual courses of the mod Language(s) of instruction and exam Weight of the module grade in the o	dule ination		German or Eng	lish	emistry"	
Compulsory entrance requirements Recommended participation require and/or individual courses of the mod Language(s) of instruction and exam Weight of the module grade in the o Frequency of module offer	dule iination overall grad		German or Eng Not graded	lish	emistry"	
Compulsory entrance requirements Recommended participation require and/or individual courses of the mod Language(s) of instruction and exam	dule iination overall grad		German or Eng Not graded	lish nmer term	emistry"	
Compulsory entrance requirements Recommended participation require and/or individual courses of the mod Language(s) of instruction and exam Weight of the module grade in the o Frequency of module offer Reasons for compulsory attendance	dule ination overall grad	de	German or Eng Not graded Only in the sun	lish nmer term Pol Besenius	emistry"	



Module MMPC	Moder	n Methods of Ph	ysical Chem	istry	[Modul-H	(ennnummer ]	
Mandatory or elective Module	Elective				-		
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	1 Semester					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lecture "Modern Methods of Physical Chemistry"	L	1 o. 2 (1 o. 2)	Ρ	3	103,5h	4,5	
b) Supporting exercise to a)	E	1 o. 2 (1 o. 2)	Р	1	34,5 h	1,5	
In order to complete the module, y	ou have to	fulfil the following	requirements:				
Compulsory Attendance							
Active participation	Accordin	g to § 5 para. 3					
Coursework					0		
Module examination	Usually w	ritten exam (120 m	in), alternatively	/ oral exam (3	0 min)		
Qualification Goals, learning outcom	ne, compe	tences					
the module, students physical chemistry, un should be able to sele corresponding measu <b>Contents</b> a) Basics of modern microscopic me Imaging microscopy meth Current topics in modern Microscopy methods for t Modern methods for the of b) In-depth or supplementary topics	thoderstand to ect the apprivation of the the the the thods with ods (confoce molecular s he analysis characterisa	he basics and be ab opriate methods fo ca in order to succes examples from thei al microscopy, scan pectroscopy, e.g. sii of dynamic process ition of molecular p	le to name poss r different expe ssfully get to the r field of applica ning probe mice ngle molecule s es and intermol hysico-chemica	ible areas of a rimental ques bottom of ne ation. Topics a roscopy, electro pectroscopy Fl ecular interact parameters (	pplication. The tions and inter w phenomena re for examples ron microscopy RET tions (FRAP) NanoSPR)	students pret the	
Compulsory entrance requirements	;						
Recommended participation requir		or the module					
and/or individual courses of the mo	odule						
and/or individual courses of the mo			German or Eng	lish			
and/or individual courses of the mo	nination	de	German or Eng Not graded	lish			
and/or individual courses of the mo Language(s) of instruction and exar	nination	de	-	lish			
and/or individual courses of the mo Language(s) of instruction and exar Weight of the module grade in the	nination overall grad	de	Not graded	lish			
and/or individual courses of the mo Language(s) of instruction and exar Weight of the module grade in the Frequency of module offer	nination overall grad	de	Not graded				
and/or individual courses of the mo Language(s) of instruction and exar Weight of the module grade in the Frequency of module offer Reasons for compulsory attendance	nination overall grad		Not graded Every term	erald Hinze			

Module MMPCP	,							
	and Mi	croscopy						
Mandatory or elective Module	Elective							
Creditpoints (LP) and workload	6 LP = 18	0 h						
Module duration (according to course plan)	1 Semes	1 Semester						
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints		
a) Practical Course "Modern Methods of Spectroscopy and Microscopy"	Apr	1 o. 2 (1 o. 2)	М	3	103,5 h	4,5		
b) Supporting seminar to a)	S	1 o. 2 (1 o. 2)	М	1	34,5 h	1,5		
In order to complete the module,	ou have to	fulfil the following	requirements:			•		
Compulsory Attendance	APr				2			
Active participation	According	g to § 5 para. 3			U.			
Coursework								
Module examination								
Qualification Goals, learning outco	me. compe	tences						
critically assessed. Contents 6-8 practical experiments from the	field of exp		•					
<ul> <li>confocal fluorescence</li> </ul>	scence and e microscop	electronic energy tra	ansfer	rried out. Exar	nples include			
	scence and e microscop nicroscopy n microscop nocrystals ecovery afte	electronic energy tra y and single molecu y er photobleaching)	ansfer Ile microscopy					
<ul> <li>confocal fluorescence</li> <li>scanning tunneling n</li> <li>light microscopy</li> <li>transmission electro</li> <li>Synthesis of CdSe na</li> <li>FRAP (fluorescence n</li> </ul>	scence and e microscopy nicroscopy n microscop nocrystals ecovery afte chosen fro	electronic energy tra y and single molecu y er photobleaching)	ansfer Ile microscopy					
<ul> <li>confocal fluorescence</li> <li>scanning tunneling n</li> <li>light microscopy</li> <li>transmission electro</li> <li>Synthesis of CdSe na</li> <li>FRAP (fluorescence r</li> <li>Topics for the oral presentation are</li> </ul>	scence and e microscopy n microscopy nocrystals ecovery afte chosen fro s rement(s) fo	electronic energy tra y and single molecu y er photobleaching) m the field of practio	ansfer Ile microscopy					
<ul> <li>confocal fluorescence</li> <li>scanning tunneling n</li> <li>light microscopy</li> <li>transmission electro</li> <li>Synthesis of CdSe na</li> <li>FRAP (fluorescence r</li> <li>Topics for the oral presentation are</li> <li>Compulsory entrance requirement</li> <li>Recommended participation requi</li> </ul>	scence and e microscopy nicroscopy nocrystals ecovery afte chosen fro s rement(s) fo odule	electronic energy tra y and single molecu y er photobleaching) m the field of practio	ansfer Ile microscopy	and related a				
<ul> <li>confocal fluorescence</li> <li>scanning tunneling n</li> <li>light microscopy</li> <li>transmission electro</li> <li>Synthesis of CdSe na</li> <li>FRAP (fluorescence r</li> <li>Topics for the oral presentation are</li> <li>Compulsory entrance requirement</li> <li>Recommended participation requiand/or individual courses of the m</li> </ul>	scence and e microscopy n microscopy nocrystals ecovery afte chosen fro s rement(s) fo odule mination	electronic energy tra y and single molecu y er photobleaching) m the field of praction or the module	ansfer Ile microscopy cal experiments	and related a				
<ul> <li>confocal fluorescence</li> <li>scanning tunneling n</li> <li>light microscopy</li> <li>transmission electro</li> <li>Synthesis of CdSe na</li> <li>FRAP (fluorescence r</li> <li>Topics for the oral presentation are</li> <li>Compulsory entrance requirement</li> <li>Recommended participation requiand/or individual courses of the m</li> <li>Language(s) of instruction and exa</li> </ul>	scence and e microscopy n microscopy nocrystals ecovery afte chosen fro s rement(s) fo odule mination	electronic energy tra y and single molecu y er photobleaching) m the field of praction or the module	ansfer Ile microscopy cal experiments German or Eng	and related a				
<ul> <li>confocal fluorescence</li> <li>scanning tunneling n</li> <li>light microscopy</li> <li>transmission electro</li> <li>Synthesis of CdSe na</li> <li>FRAP (fluorescence r</li> <li>Topics for the oral presentation are</li> <li>Compulsory entrance requirement</li> <li>Recommended participation requi</li> <li>and/or individual courses of the m</li> <li>Language(s) of instruction and exa</li> <li>Weight of the module grade in the</li> </ul>	scence and e microscopy n microscopy nocrystals ecovery afte chosen fro s rement(s) fo odule mination overall gra	electronic energy tra y and single molecu y er photobleaching) m the field of praction or the module	ansfer ile microscopy cal experiments German or Eng Not graded	and related a	reas.	ctical Course		
<ul> <li>confocal fluorescence</li> <li>scanning tunneling n</li> <li>light microscopy</li> <li>transmission electro</li> <li>Synthesis of CdSe na</li> <li>FRAP (fluorescence n</li> <li>Topics for the oral presentation are</li> <li>Compulsory entrance requirement</li> <li>Recommended participation requiand/or individual courses of the m</li> <li>Language(s) of instruction and exa</li> <li>Weight of the module grade in the</li> <li>Frequency of module offer</li> </ul>	scence and e microscopy nicroscopy nocrystals ecovery after chosen fro s rement(s) fro odule mination overall gra	electronic energy tra y and single molecu y er photobleaching) m the field of praction or the module	ansfer Ile microscopy cal experiments German or Eng Not graded Every term	and related a lish ochSchG § 26	reas.	ctical Course		
<ul> <li>confocal fluorescence</li> <li>scanning tunneling n</li> <li>light microscopy</li> <li>transmission electro</li> <li>Synthesis of CdSe na</li> <li>FRAP (fluorescence r</li> <li>Topics for the oral presentation are</li> <li>Compulsory entrance requirement</li> <li>Recommended participation requi</li> <li>and/or individual courses of the m</li> <li>Language(s) of instruction and exa</li> <li>Weight of the module grade in the</li> <li>Frequency of module offer</li> <li>Reasons for compulsory attendance</li> </ul>	scence and e microscopy n microscopy nocrystals ecovery afte chosen fro s rement(s) fo odule mination overall gra	electronic energy tra y and single molecu y er photobleaching) m the field of praction or the module de	ansfer ile microscopy cal experiments German or Eng Not graded Every term According to H	and related a lish ochSchG § 26 erald Hinze	reas.	ctical Course		

Module KC	Introdu	<mark>ction in Nuclea</mark> i	<mark>r Chemistry</mark>		[Modul-K	ennnummer ]	
Mandatory or elective Module	Elective				<u>_</u>		
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration     1 Semester       (according to course plan)     1							
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lecture "Introduction in Nuclear Chemistry"	L	1 o. 2 (1 o. 2)	М	2	69 h	3	
b) Supporting exercise to a)	Е	1 o. 2 (1 o. 2)	М	1	34,5 h	1,5	
c) Supporting Seminar to a)	S	1 o. 2 (1 o. 2)	М	1	34,5	1,5	
In order to complete the module, yo	ou have to	fulfil the following	requirements:	,,			
Compulsory Attendance	S						
Active participation	According	g to § 5 para. 3		.0	•		
Coursework							
Module examination	Usually w	ritten exam (120 m	in), alternatively	y oral exam (30	) min)		
Qualification Goals, learning outcon	ne, compe	tences					
Contents a) History of radioactivity / structure liquid droplet model and shell model transformations, units of radioactivit radioactivity, spontaneous fission / s Annihilation, X-rays, Auger electrons measurement of nuclear radiation: d compound nuclei, heavy ion reaction b) In the exercises, exercise assignment c) Presentations will be given on topid determination; Discovery and proper environment; The tracer principle an	I / instabilit econdary c / interacti lifferent ty ns, high ene ents are ca ics that cor	ty of nuclei and nucl radionuclides / prim conversions: electro on with matter: pho pes of detectors / n ergy reactions, induc lculated. nplement the lectur	lear transformation ary transformation magnetic transion toelectric effecuclear reactions ced fission.	tion principles tions: α-conve tions, convers t, Compton ef : Energetics, c	/ mathematica rsion, β-conver on electrons / fect, pair forma	l relations of sion, cluster post effects: tion /	
	id its applic		of nuclear fission	on; Natural rad	lioactivity in th	e	
and operation; Neutron activation ar fusion; Production and properties of	iences; Nuo nalysis; Nuo transurani	cations in chemistry clear medicine diagi clear fuel cycle; The	of nuclear fission and medicine; I nostics; Biologic Chernobyl and	on; Natural rad Particle accele al radiation ef Fukushima rea	lioactivity in the rators; Product fects; Nuclear r	e ion and eactor design	
and operation; Neutron activation ar fusion; Production and properties of <b>Compulsory entrance requirements</b>	iences; Nuo nalysis; Nuo transurani	ations in chemistry clear medicine diag clear fuel cycle; The um elements; Solar	of nuclear fission and medicine; I nostics; Biologic Chernobyl and	on; Natural rad Particle accele al radiation ef Fukushima rea	lioactivity in the rators; Product fects; Nuclear r	e ion and eactor design	
and operation; Neutron activation ar fusion; Production and properties of Compulsory entrance requirements Recommended participation require	iences; Nuo nalysis; Nuo transurani ement(s) fo	ations in chemistry clear medicine diag clear fuel cycle; The um elements; Solar	of nuclear fission and medicine; I nostics; Biologic Chernobyl and	on; Natural rad Particle accele al radiation ef Fukushima rea	lioactivity in the rators; Product fects; Nuclear r	e ion and eactor design	
and operation; Neutron activation ar fusion; Production and properties of Compulsory entrance requirements Recommended participation require and/or individual courses of the mo	iences; Nuo nalysis; Nuo transurani ement(s) fo dule	ations in chemistry clear medicine diag clear fuel cycle; The um elements; Solar	of nuclear fission and medicine; I nostics; Biologic Chernobyl and	on; Natural rad Particle accele al radiation ef Fukushima rea	lioactivity in the rators; Product fects; Nuclear r	e ion and eactor design	
and operation; Neutron activation ar fusion; Production and properties of Compulsory entrance requirements Recommended participation require and/or individual courses of the mo Language(s) of instruction and exam	iences; Nuc halysis; Nuc transurani ement(s) fo dule hination	cations in chemistry clear medicine diago clear fuel cycle; The um elements; Solar or the module	of nuclear fission and medicine; I nostics; Biologic Chernobyl and and atmospher	on; Natural rad Particle accele al radiation ef Fukushima rea	lioactivity in the rators; Product fects; Nuclear r	e ion and eactor design	
and operation; Neutron activation ar fusion; Production and properties of Compulsory entrance requirements Recommended participation require and/or individual courses of the mo Language(s) of instruction and exam Weight of the module grade in the c	iences; Nuc halysis; Nuc transurani ement(s) fo dule hination	cations in chemistry clear medicine diago clear fuel cycle; The um elements; Solar or the module	of nuclear fission and medicine; I nostics; Biologic Chernobyl and and atmospher German Not graded Every term	on; Natural rac Particle accele al radiation ef Fukushima rea ic neutrinos.	lioactivity in th rators; Product fects; Nuclear r actor disasters;	e ion and eactor design Nuclear	
application of radionuclides in life sci and operation; Neutron activation ar fusion; Production and properties of Compulsory entrance requirements Recommended participation require and/or individual courses of the mo Language(s) of instruction and exam Weight of the module grade in the c Frequency of module offer Reasons for compulsory attendance	iences; Nuc halysis; Nuc transurani ement(s) fo dule hination overall grad	cations in chemistry clear medicine diago clear fuel cycle; The um elements; Solar or the module	of nuclear fission and medicine; I nostics; Biologic Chernobyl and and atmospher German Not graded	on; Natural rac Particle accele al radiation ef Fukushima rea ic neutrinos. ding to § 5 para irect interactio ctical professio ives are literat	lioactivity in th rators; Product fects; Nuclear r fector disasters; 	e ion and eactor design Nuclear 	



Transferability of the module to other degree programs	Bachelor of Science Biomedicinal Chemistry, Bachelor of Science Chemistry, Bachelor of Science Geoscience, Master of Science Chemistry, Master of Science Physics
Other	<ul> <li>Recommended Literature:</li> <li>JV. Kratz, K. H. Lieser: Nuclear and Radiochemistry, Wiley-VCH, 2013</li> <li>F. Rösch: Nuclear and Radiochemistry, De Gruyter, 2014</li> <li>Vértes, S. Nagy, Z. Klencsár, R. G. Lovas, F. Rösch (Eds.), Handbook of Nuclear Chemistry, Springer 2011</li> </ul>
	noutebarantee
hormation	

Module KCP	<mark>Lab Coເ</mark>	irse Nuclear Cho	emistry 1		[Modul-I	Kennnummer ]	
Mandatory or elective Module	Elective				<u>_</u>		
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semest	er					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lab Course "Nuclear Chemistry 1"	APr	1 o. 2 (1 o. 2)	М	6	72 h	4,5	
b) Supporting Seminar to a)	S	1 o. 2 (1 o. 2)	М	1	34,5h	1,5	
In order to complete the module, yo	u have to	fulfil the following	requirements:	•		· · · ·	
Compulsory Attendance	APr, S				X	0	
Active participation	According	to § 5 para. 3					
Coursework		•					
Module examination	Oral exam	n (30 minutes, not g	raded)		0		
Qualification Goals, learning outcom	<u> </u>		· •				
time and resource mar to organise themselves Contents Production and handling of radioactiv equilibrium, interaction of radiation v emission tomography, nuclear reaction behaviour of neptunium.	s in small g ve prepara with matte	tions, measuremen r, gamma spectroso	t of alpha, beta copy, dosimetry	, gamma radia and radiation	protection, ba	sics of positron	
Compulsory entrance requirements			Module "Intro	duction in nuc	lear Chemistry	u	
Recommended participation require and/or individual courses of the mod	• •	or the module					
Language(s) of instruction and exam	ination		German				
Weight of the module grade in the o	verall grad	le	Not graded				
Frequency of module offer			Every term				
Reasons for compulsory attendance			In accordance with HochSchG § 26 para. 2 (7), practical course; seminar accompanying practical course in accordance with § 5 para. 5: discussion of safety-relevant details of and discussion of practical course experiments.				
Person responsible for the module			UnivProf. Tho				
Transferability of the module to othe	er degree	programs	Bachelor of Science Biomedical Chemistry, Bachelor of Science Chemistry, Bachelor of Science Geoscience, Master of Science Chemistry, Master of Science Physics				
Other			und I • W. St • HG.	offmann, K. H. Radiochemie, <sup>v</sup> tolz: Radioakti Vogt, H. Schu	Lieser: Methoo VCH 1991 vität, Teubner, Itz: Grundzüge enschutzes, Hau	2005 des	



Mandatory or elective Module	1		Principles of Quantum Chemistry				
	Elective						
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semester						
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Lecture "Principles of Quantum Chemistry"	L	1 o. 3 (2)	м	3	103 <i>,</i> 5 h	4,5	
<ul><li>b) Supporting exercise to a)</li></ul>	Е	1 o. 3 (2)	М	1	34,5 h	1,5	
n order to complete the module, yo	ou have to	fulfil the following	requirements:			•	
Compulsory Attendance							
Active participation	According	g to § 5 para. 3					
Coursework					0		
Module examination	Usually w	ritten exam (120 m	in), alternatively	/ oral exam (30	) min)		
They acquire a profound understand In the context of quantum chemistry how the equations are solved and ar Contents Molecular orbitals and Hartree-Fock theory (g Self-consistent field m Basis set representation Implementation of HF Molecular properties	. They will e able to d d multi-eled general ide tethod for on and Roo -SCF and p	be able to perform esign a correspondi ctron wave function a, detailed derivation solving the HF equa othaan-Hall equation erformance of corre	the derivation c ng computer pr on of the correst tions ns esponding calcu	of the correspo ogram. ponding equat	onding equation		
Recommended participation require		or the module					
and/or individual courses of the mo							
Language(s) of instruction and exam			German or English				
Weight of the module grade in the c	overall grad	de	Not graded				
Frequency of module offer			Only in the win	ter term			
Reasons for compulsory attendance Person responsible for the module			UnivProf. Dr.	lürgen Gauß			
croon responsible for the module				-			
Transferability of the module to oth	er degree	nrograms	Master of Scier	ICP ( hpmictry			

Module PQC	Programming in Quantum Chemistry			[Modul-k	[Modul-Kennnummer ]	
Mandatory or elective Module	Elective					
Creditpoints (LP) and workload	6 LP = 180 h					
Module duration (according to course plan)	1 Semester					
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints
a) Practical Course "Programming in Quantum Chemistry"	Apr	1 o. 3 (2)	М	3	103,5 h	4,5
b) Supporting Seminar to a)	S	1 o. 3 (2)	М	1	34,5 h	1,5
In order to complete the module, you	have to	fulfil the following	requirements:			
Compulsory Attendance	Apr, S					
Active participation	According	to § 5 para. 3				
Coursework					0	
Module examination						
Qualification Goals, learning outcome	e, compet	ences				
and critically discuss the Contents Basics of programming Planning and conceptio Implementation of qua	on of a cor		computer prog	ramme		
Compulsory entrance requirements			Module "Principles of Quantum Chemistry"			
Recommended participation requirer and/or individual courses of the mod		or the module				
Language(s) of instruction and exami	nation		German or English			
Weight of the module grade in the ov	verall grad	le	Not graded			
Frequency of module offer			Only in the winter term in the semester break			
Reasons for compulsory attendance			According to HochSchG § 26 Para. 2 (7), internship; internship-accompanying upper seminar according to § 5 Para. 5: Discussion of the tasks to be carried out or carried out in the internship with the help of licensed programmes on computers within the working group.			
Person responsible for the module			UnivProf. Dr. Jürgen Gauß			
Transferability of the module to other degree programs			Master of Science Biomedicinal Chemistry			
Transferability of the module to othe	ucgice	ргодланиз			iai circinistry	



Mandatory or elective Module	Practical Computational Chemistry [Modul-Kennnummer]				(ennnummer ]		
	Elective						
Creditpoints (LP) and workload	6 LP = 18	0 h					
Module duration (according to course plan)	1 Semester						
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoints	
a) Practical Course Computer Chemistry	APr	2 (1 o. 3)	М	3	103,5 h	4,5	
b) Supporting Seminar to a)	S	2 (1 o. 3)	М	1	34,5 h	1,5	
In order to complete the module, you	u have to	fulfil the following	requirements:				
Compulsory Attendance	APr, S						
Active participation	According	to § 5 para. 3					
Coursework					0		
Module examination							
Qualification Goals, learning outcome	e, compet	ences					
methods from the field Contents Carrying out 2-4 exemplary experimer the fields of AC, OC, PC, KC and/or bio	nts in whi	ch chemical issues a	ire investigated	from a combin	nation of exper		
Compulsory entrance requirements			Madulas "Cant	omporary Tor	ice of Quantum	Chomistry"	
Recommended participation requirer and/or individual courses of the mod		or the module	Modules "Contemporary Topics of Quantum Chemistry" and "Principles of Theoretical Chemistry"				
Language(s) of instruction and exami	ination		German or English				
Weight of the module grade in the ov	verall grad	le	Not graded				
Frequency of module offer			Only in the summer term in the lecture-free period				
Reasons for compulsory attendance		According to HochSchG § 26 Para. 2 (7), practical course; internship-accompanying upper seminar according to § 5 Para. 5: Discussion of the tasks to be carried out or carried out in the internship with the help of licensed programmes on computers within the working group.					
Person responsible for the module			UnivProf. Dr. Jürgen Gauß				
Person responsible for the module	Transferability of the module to other degree programs			Master of Science Chemistry			
			Block practical course				

Module MTTC Contemporary Topics of Theoretical Chemistry				[Modul-H	[Modul-Kennnummer ]	
Mandatory or elective Module	Elective					
Creditpoints (LP) and workload	6 LP = 18	0 h				
Module duration (according to course plan)	1 Semest	1 Semester				
Courses/ Learning formats	Туре	Regular term when starting in Winter term (Summer term)	Mandatory/ elective	Contact Time (SWH)	Self Study	Creditpoint
a) Lecture "Contemporary Topics of Theoretical Chemistry"	L	2 (1 o. 3)	М	3	103,5 h	4,5
<ul><li>b) Supporting exercise to a)</li></ul>	E	2 (1 o. 3)	М	1	34,5 h	1,5
In order to complete the module, yo	u have to	fulfil the following	requirements:			
Compulsory Attendance						
Active participation	According	g to § 5 para. 3			0	
Coursework						
Module examination	Usually w	ritten exam (120 m	in), alternativel	y oral exam (3	0 min)	
Qualification Goals, learning outcom	e, compe	tences				
They have developed t     Chemistry in Practice". Contents		tical foundations fo	r the calculation	ns required in t	the module "Co	omputational
Advanced quantum chemic Theoretical description of many-parti			tion, electron co	orrelation		
Compulsory entrance requirements						
Recommended participation require and/or individual courses of the mod		or the module				
Language(s) of instruction and exam	ination		German or English			
Weight of the module grade in the o	verall gra	de	Not graded			
Frequency of module offer			Only in the summer term			
Reasons for compulsory attendance						
Person responsible for the module			UnivProf. Dr. Jürgen Gauß			
Transferability of the module to other degree programs			Master of Science Chemistry			
Other						

## Remarks

Depending on the type of course, different conversion factors are used to calculate a certain number of ECTS.

In general:

1 ECTS corresponds to 30h total workload (time hours),

1 corresponds to 10.5h attendance time per semester (14 weeks à 0.75h)

xe

Contact time (SWH)	1	2	3	4
Total attendance time	10,5h	21h	31 <i>,</i> 5h	42h

### Lectures and/or Exercises

A factor of 1.5 is applied, i.e. 2 lectures or exercises correspond to 3 ECTS.

4,5 ECTS	3 (e.g. 2L+1E), 31,5h attendance time, 103.5h self-study, 135h total workload
6,0 ECTS	4 (e.g. 3L+1E), 42h attendance time, 138h self-study, 180h total workload
7,5 ECTS	5 (e.g. 3L+2E), 52,5h attendance time, 172,5h self-study, 225h total workload

### **Practical Courses**

A factor of 0.50 or 0.75 or 1.00 is applied, depending on the extent of preparation and follow-up, e.g. with reports, ...

6,0 ECTS	Factor 0,50	12, 126h attendance time, 54h self-study, 180h total workload
		e.g. 10 weeks of 12,6h
7,5 ECTS	Factor 0,50	15, 157,5h attendance time, 67,5h self-study, 225h total workload
		e.g. 10 weeks of 15h
7,5 ECTS	Factor 0,75	10, 105h attendance time, 120h self-study, 225h total workload
		e.g. 10 weeks of 10,5h
6,0 ECTS	Factor 1,00	6, 63h attendance time, 117h self-study, 180h total workload
		e.g. 10 weeks of 6,3h

#### Seminars

A factor of 1.0 or 1.5 is applied, depending on the amount of preparation and follow-up, e.g. with lectures, new learning material, ...

1,0 ECTS	Factor 1,0	1, 10,5h attendance time, 19,5h self-study, 30h total workload
2,0 ECTS	Factor 1,0	2, 21h attendance time, 39h self-study, 60h total workload
1,5 ECTS	Factor 1,5	1, 10,5h attendance time, 34,5h self-study, 45h total workload
3,0 ECTS	Factor 1,5	2, 21h attendance time, 69h self-study, 90h total workload

Module Handbook	Version	As of	JG
Master of Science Biomedical Chemistry	2.0	14.03.2024	JGIO

# Abbreviations

Abbreviation	Meaning	]
ВМС	Biomedical Chemistry	-
e.g.	For example	
ECTS / CP(LP)	European Credit Transfer System / Credit Point	
IUPAC	International Union of Pure and Applied Chemistry	
SWH(SWS)	Hours per Semester Week	
S	Seminar	0
E	Exercise	
Apr	Advanced Practical Course	
L	Lecture	
hormati	on	